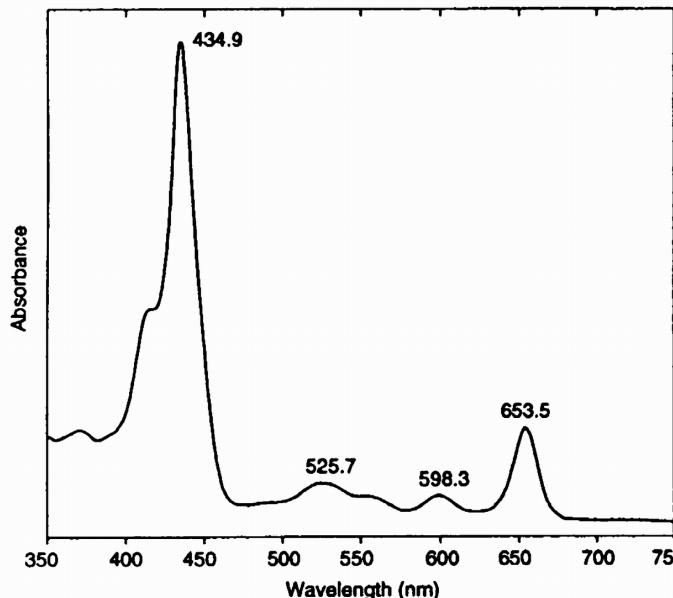


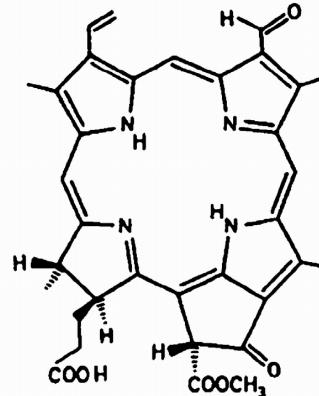
# Pheophorbide *b*

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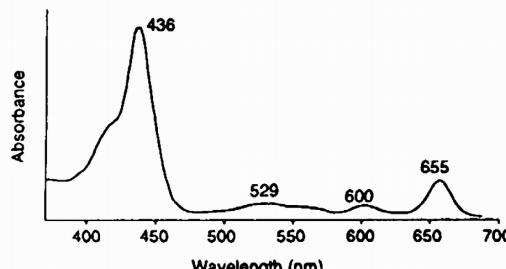
Standard spectrum in reference solvent: acetone (100%)



## Molecular structure



## Diode array spectrum in SCOR eluant



## HPLC: Pheophorbide *b*

Prepared from acidified chlorophyllide *b*

## Property

## Data

Name: (Trivial)  
(IUPAC)

**Pheophorbide *b***  
Trivial name sufficient, see Hynninen (1991)

SCOR abbreviation:

Phide *b*

Occurrence:

Terrestrial plant detritus, sediments

Colour:

Grey (red fluorescence) on TLC; yellow-grey (concentrated solution)

Molecular formula:

C<sub>35</sub>H<sub>34</sub>N<sub>4</sub>O<sub>6</sub>

Molecular weight:

606.67

Specific extinction coefficient:

$\alpha$  (l g<sup>-1</sup> cm<sup>-1</sup>)

46.37 (at 657 nm in 90% acetone)

Calculated from Lorenzen & Jeffrey (1980)

Molar extinction coefficient:

$\epsilon$  (l mol<sup>-1</sup> cm<sup>-1</sup>)

28.1 x 10<sup>3</sup> (at 657 nm in 90% acetone)

Calculated from  $\alpha$  above

UV-vis spectra:

Solvent	Absorbance maxima ( nm)				Band ratio*	Reference
100% Acetone	434.9	525.7	598.3	653.5	4.64	Repeta (unpub. data)
Diethyl ether	432			654	4.81	Brown (1968)
HPLC Eluant	436	529	600	655	5.23	SCOR WG 78: Wright <i>et al.</i> (1991) method

Fluorescence spectra:

\*Soret (blue maximum): red ratio

Solvent	Excitation ( nm)	Emission ( nm)	Reference
No data available			

Alteration products:

Origin: Acidification of chlorophyllide *b*

Additional reference(s):

Scheer (1991)