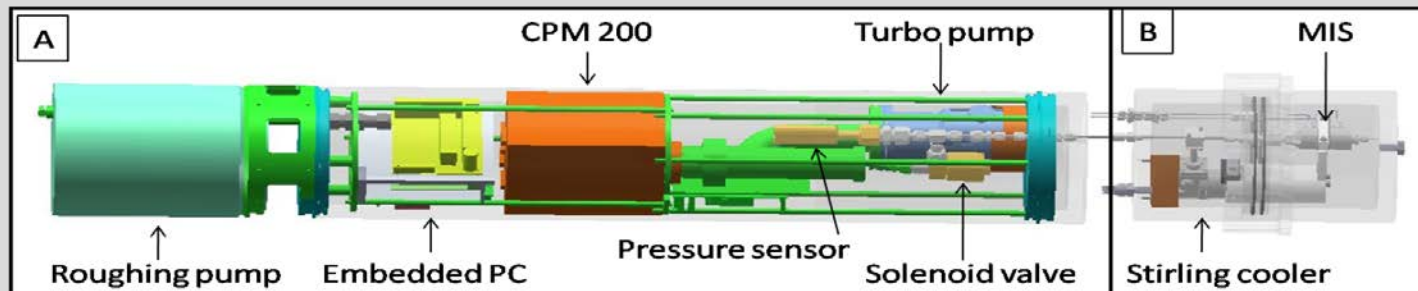


In Situ Mass Spectrometry in Marine Science: Distribution and Fate of Methane Released from Submarine Sources

Torben Gentz & Michael Schlüter

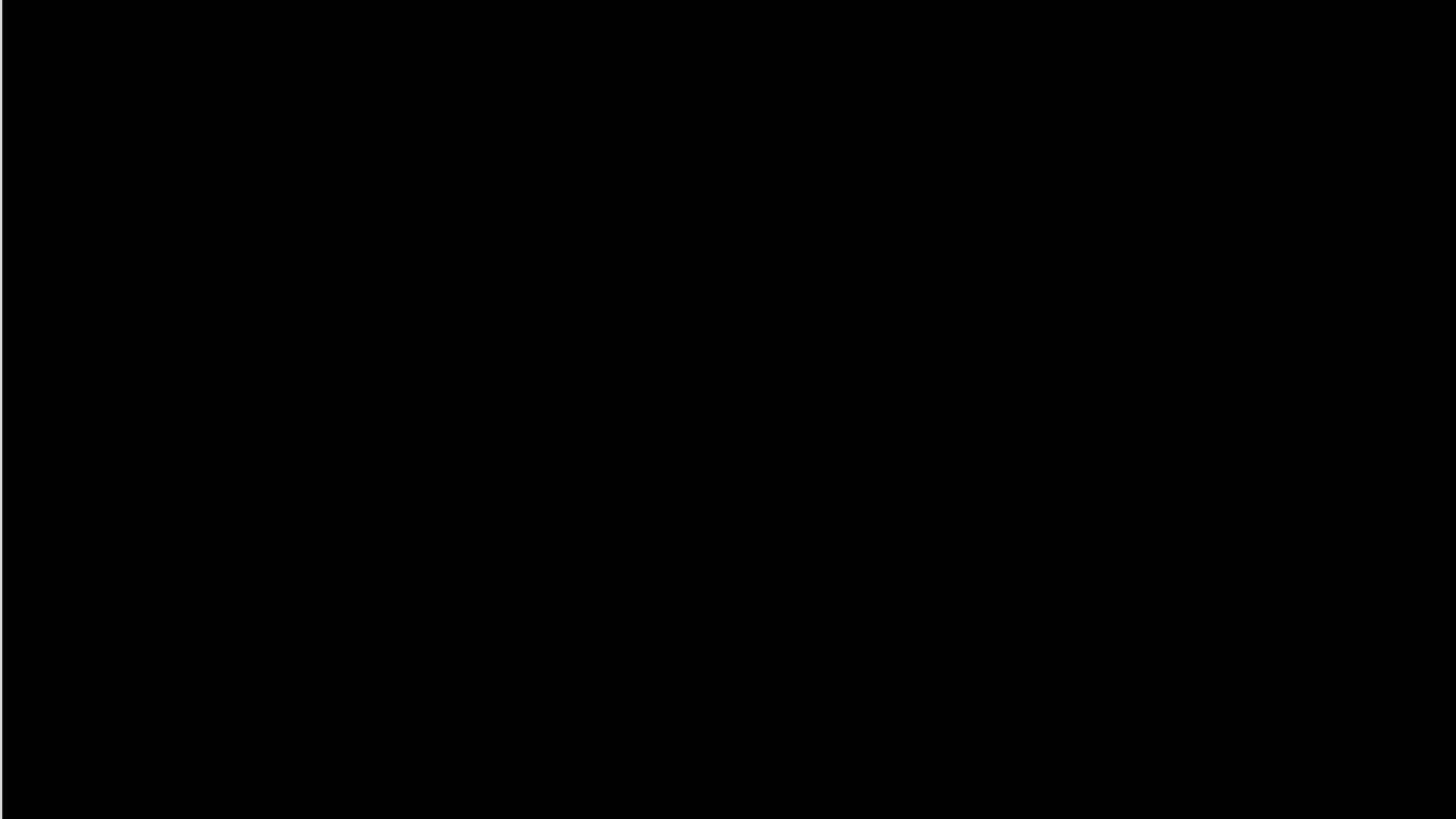
Alfred Wegener Institute for Polar and Marine Research
Bremerhaven, Germany

Torben.Gentz@awi.de



MARINE SCIENCE IS HARSH ENVIRONMENT !





Heincke 362

Submarine gas seeps

WORLDWIDE DISTRIBUTION OF SUBMARINE METHANE RELEASE

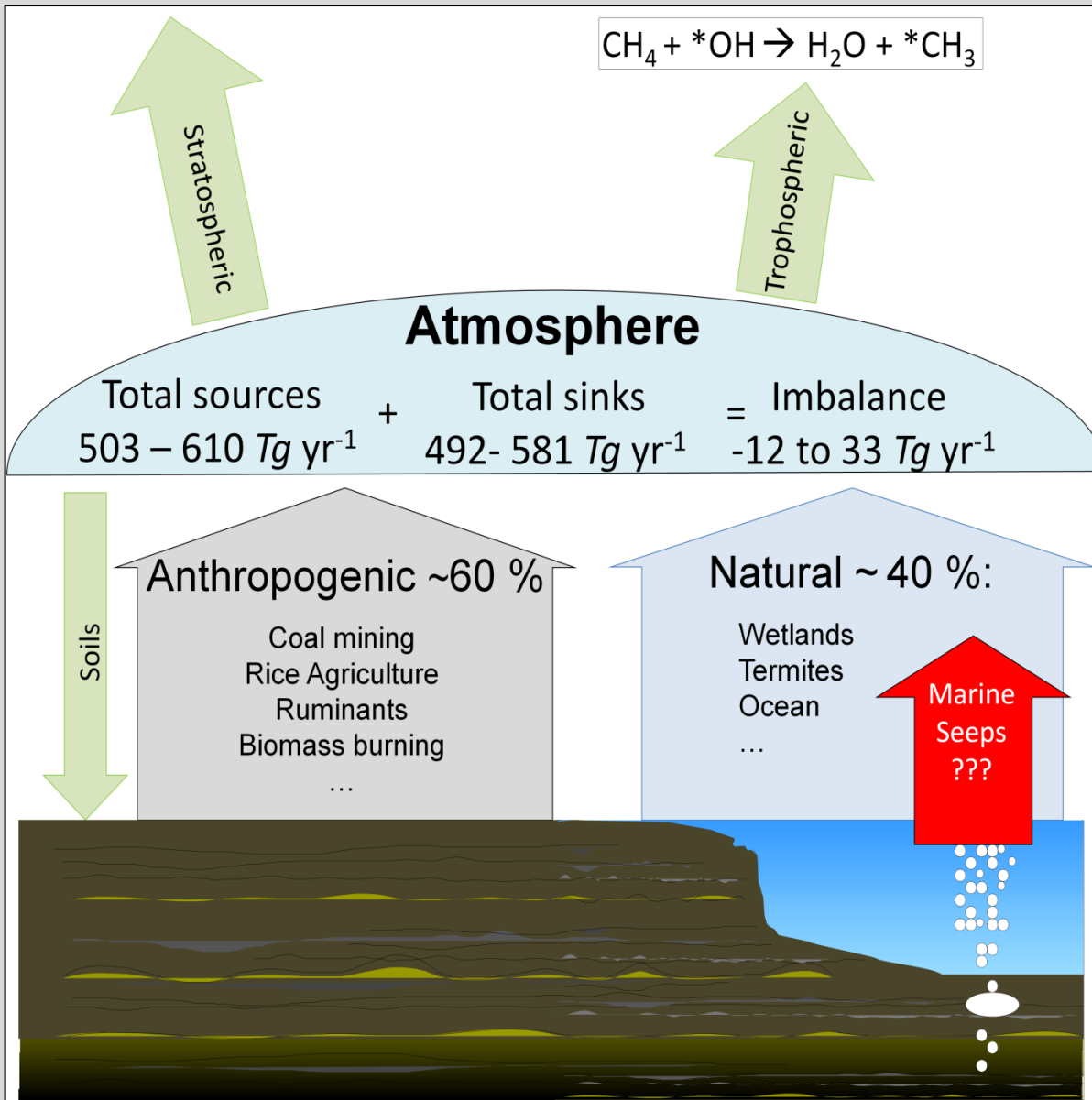
Free gas (Fleischer et al. 2001)

Pockmarks (Hovland et al. 2002)

Mud volcanoes (Milkov 2000)

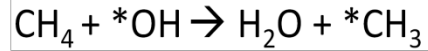
Gas hydrates (Kvenvolden et al. 2001)

GLOBAL RELEVANCE OF METHANE



according to [Intergovernmental Panel on Climate Change \(IPCC, 2007\)](#)

GLOBAL RELEVANCE OF METHANE



Stratospheric

Tropospheric

Atmosphere

Total sources + Total sinks = Imbalance
 $503 - 610 \text{ Tg yr}^{-1} + 492 - 581 \text{ Tg yr}^{-1} = -12 \text{ to } 33 \text{ Tg yr}^{-1}$

Soils

Anthropogenic ~60 %

Coal mining
 Rice Agriculture
 Ruminants
 Biomass burning
 ...

Natural ~40 %:

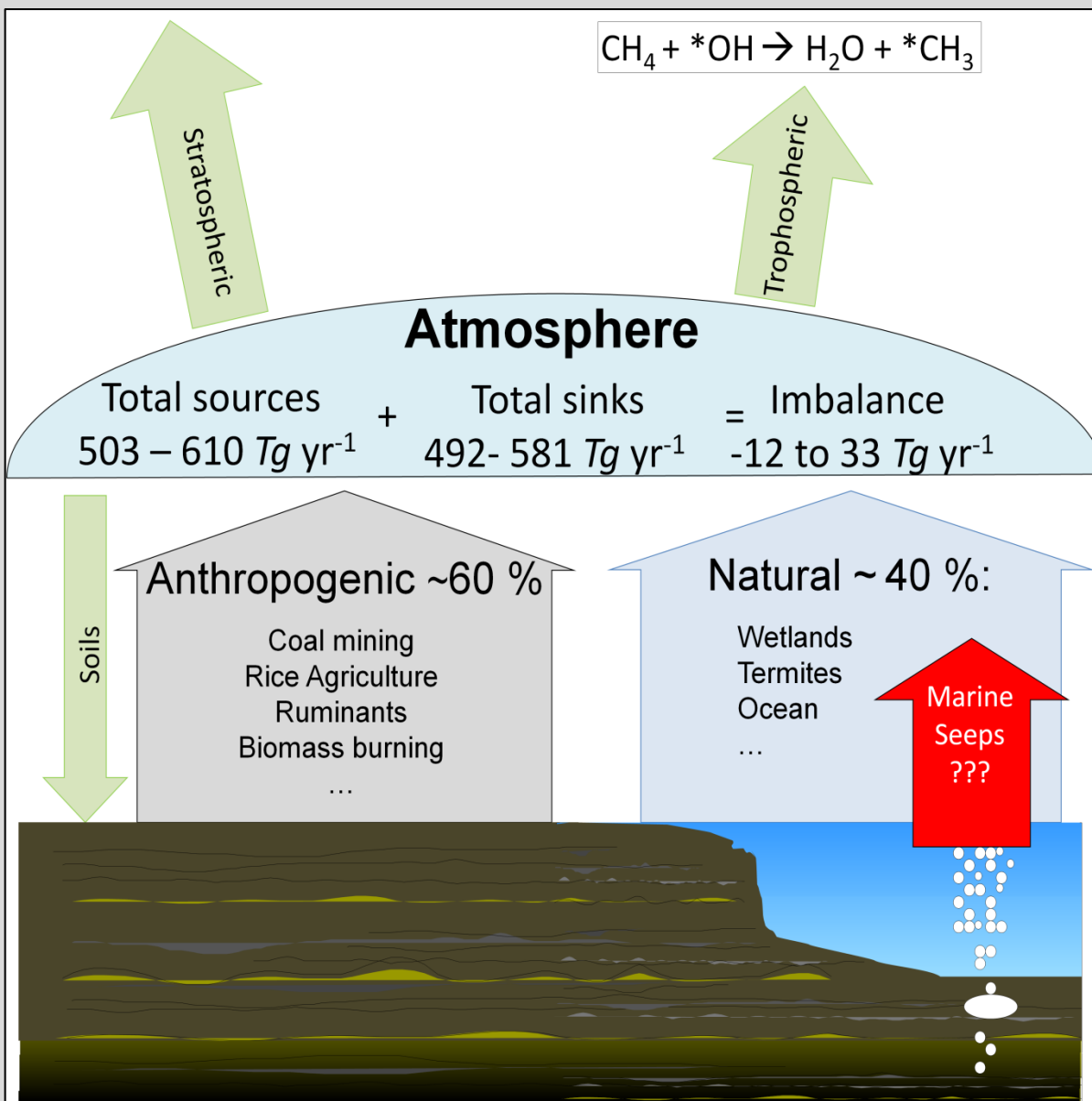
Wetlands
 Termites
 Ocean
 ...

Marine
 Seeps
 ???

according to IPCC (2007)

The average atmospheric concentration of methane has increased by 151 % since year 1750 (Houghton 2001).

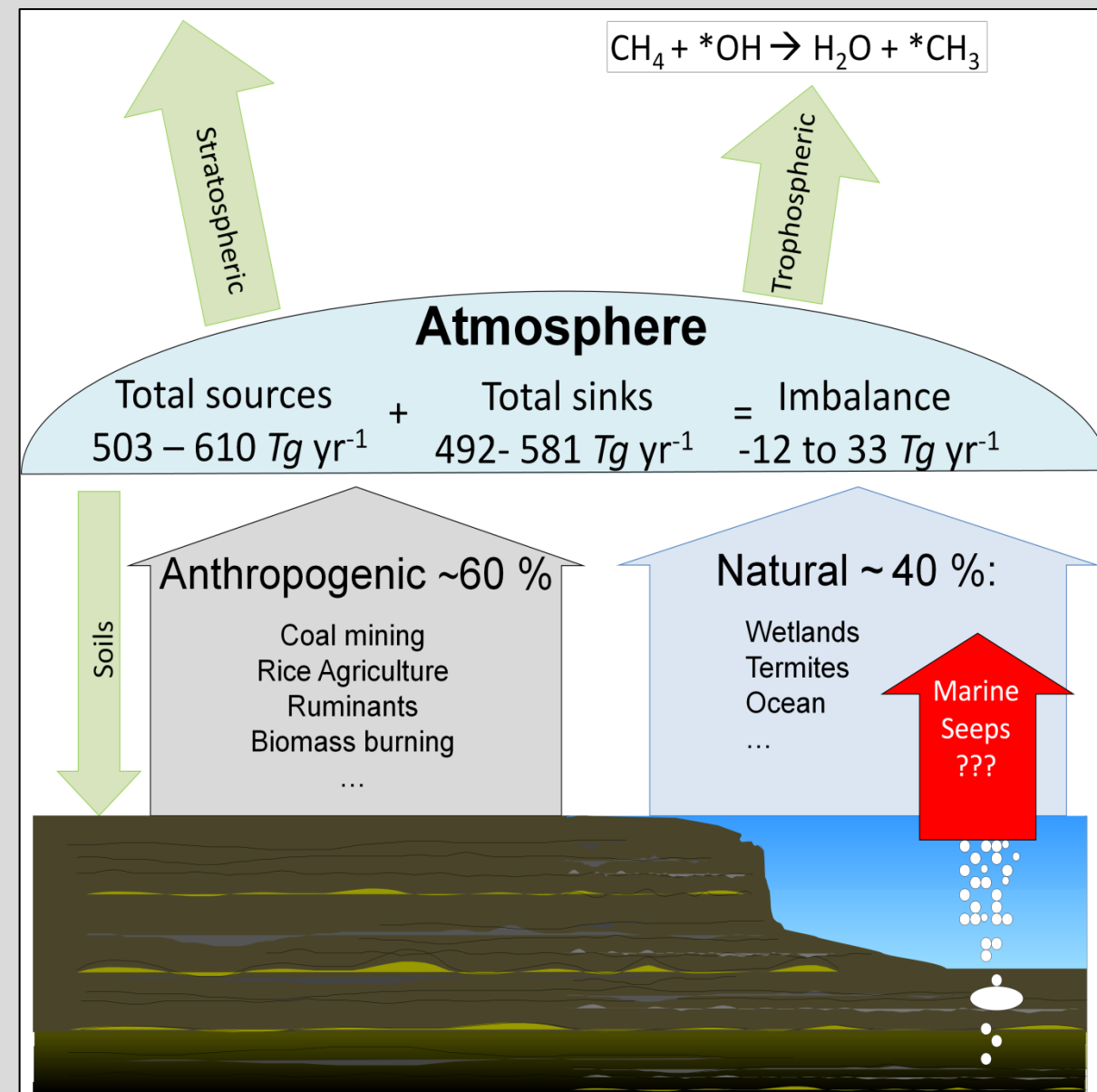
GLOBAL RELEVANCE OF METHANE



The average atmospheric concentration of methane has increased by 151 % since year 1750 (Houghton 2001).

CH_4 acts beside CO_2 and water vapour as a greenhouse gas (Houghton 2001).

GLOBAL RELEVANCE OF METHANE

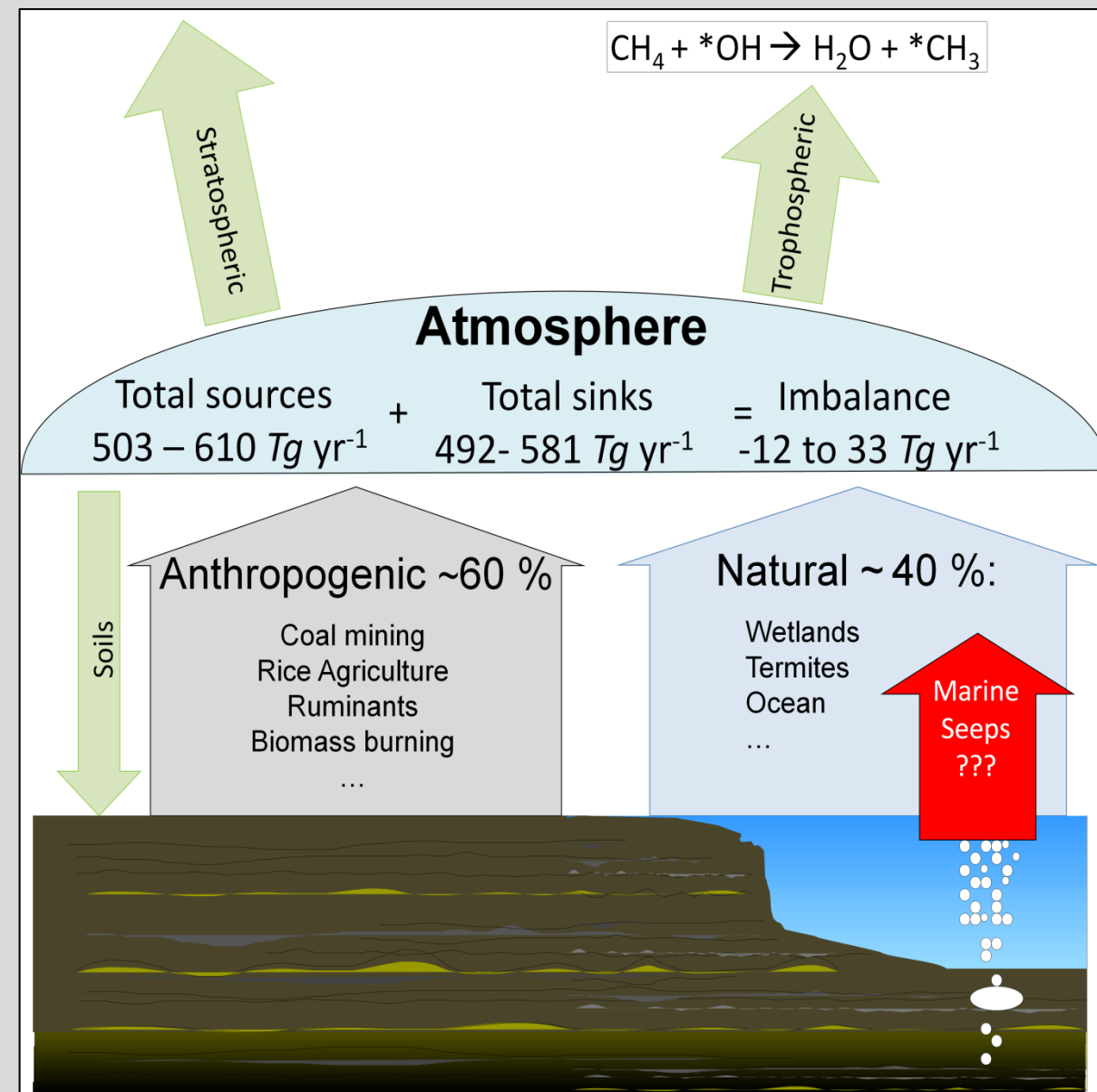


The average atmospheric concentration of methane has increased by 151 % since year 1750 (Houghton 2001).

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On a 100 year timescale the global warming potential (GWP) of CH_4 is 20 – 40 times higher than of CO_2 (Shindell 2009).

GLOBAL RELEVANCE OF METHANE



according to IPCC (2007)

The average atmospheric concentration of methane has increased by 151 % since year 1750 (Houghton 2001).

CH_4 acts beside CO_2 and water vapour as a greenhouse gas (Houghton 2001).

On a 100 year timescale the global warming potential (GWP) of CH_4 is 20 – 40 times higher than of CO_2 (Shindell 2009).

CH_4 represents the second largest contribution (about 15 %) to historical warming after CO_2 (Shindell et.al. 2009).

GLOBAL RELEVANCE OF SUBMARINE SOURCES

Present estimations: 8 - 65 Tg CH₄ yr⁻¹ are released into the ocean and 0.4 – 48 Tg CH₄ yr⁻¹ reach the atmosphere which is up to 9 % of the total methane emission (Hovland et al. 1993; Judd and Hovland 2007; Judd 2004; Judd et al. 2002; Kvenvolden and Rogers 2005).

Future Scenarios induced by global warming:

Thawing of permafrost (e.g. Shakhova et al. 2010)

Destabilization of gas hydrates (e.g. Jung and Vogt 2004; Mienert et al. 2005; Ruppel 2011)

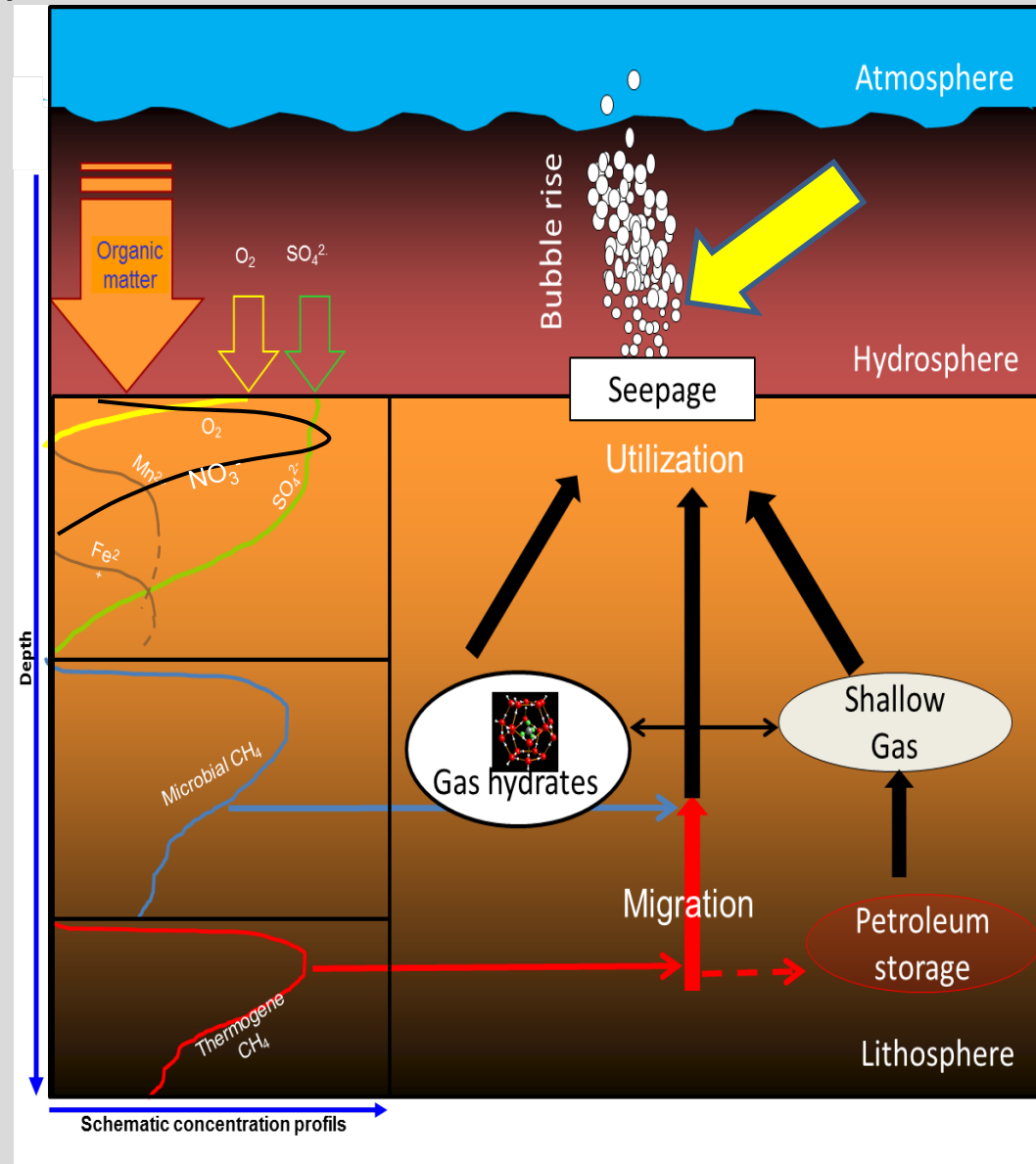
Free gas (Fleischer et al. 2001)

Pockmarks (Hovland et al. 2002)

Mud volcanoes (Milkov 2000)

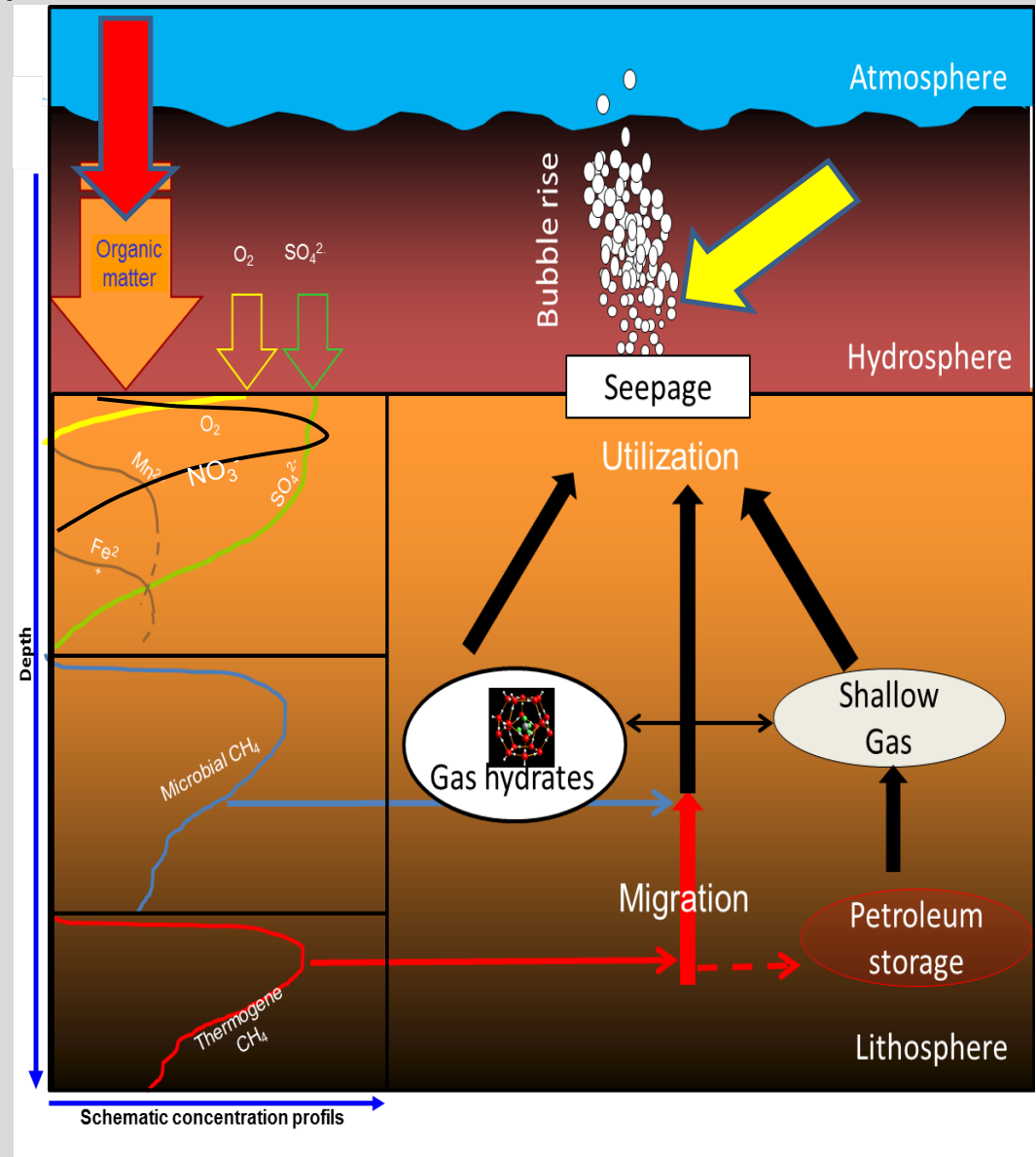
Gas hydrates (Kvenvolden et al. 2001)

WHAT ARE SUBMARINE GAS SEEPS?



Schematic view of the formation (modified after Froelich et al. 1979) and the subsequent pathways of methane in the sediment (modified after Judd 2004). Crystallographic image of gas hydrates after Bohrmann and Torres (2006)

WHAT ARE SUBMARINE GAS SEEPS?



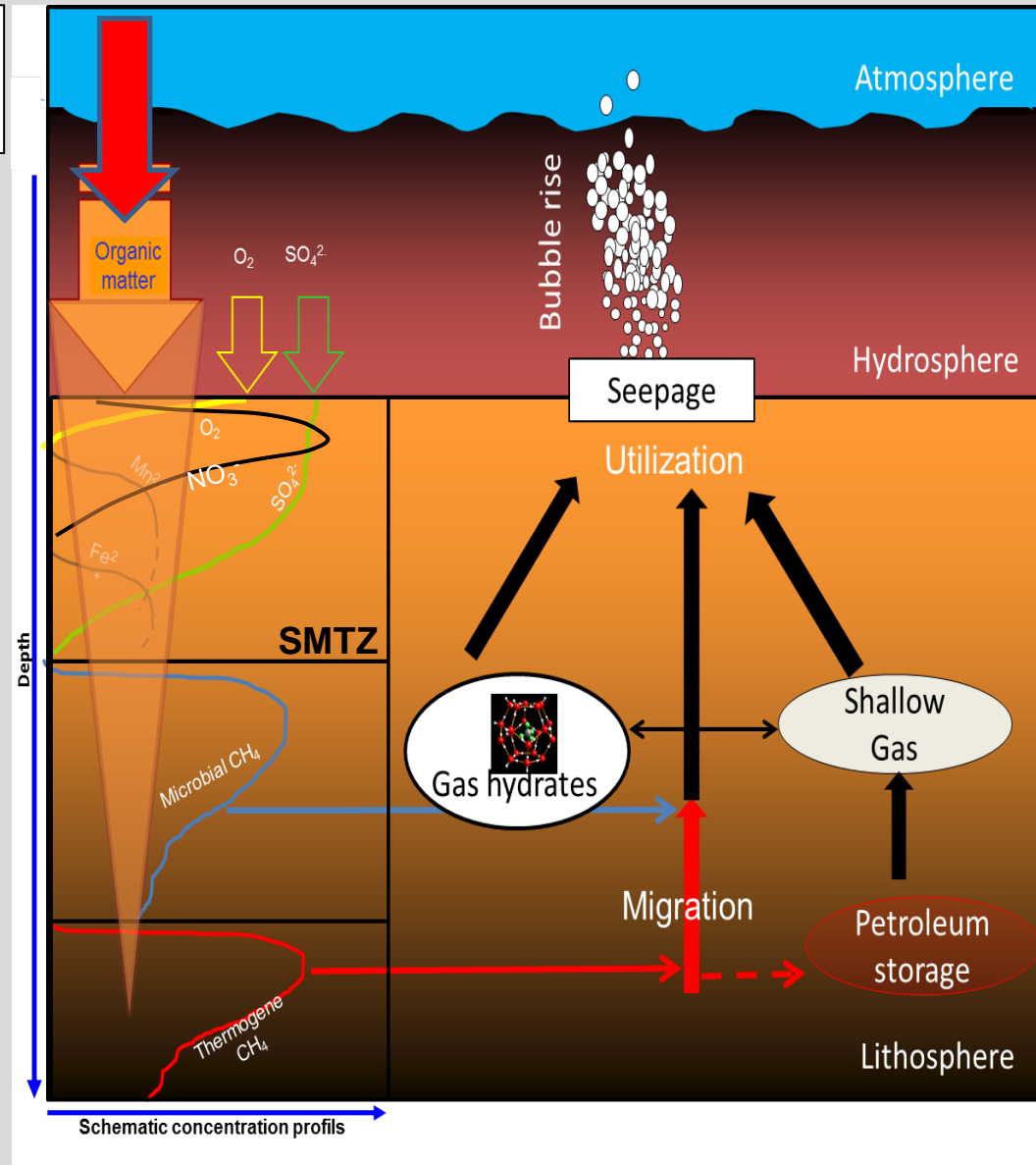
Schematic view of the formation (modified after Froelich et al. 1979) and the subsequent pathways of methane in the sediment (modified after Judd 2004). Crystallographic image of gas hydrates after Bohrmann and Torres (2006)

Formation of methane by degradation of organic matter

Aerobic respiration
Nitrate reduction
Manganese oxide reduction
Iron oxide reduction

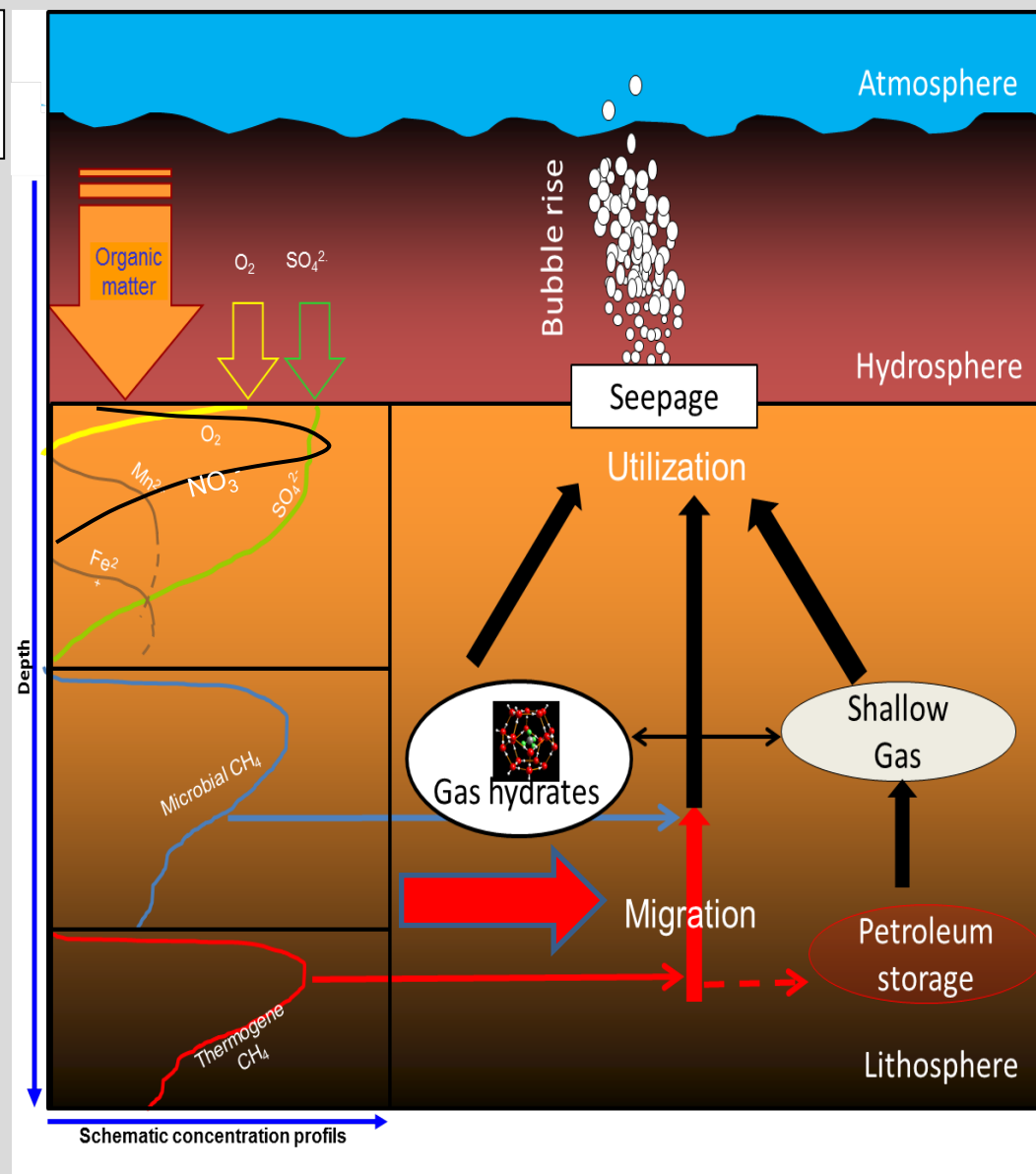
Microbial formation of methane

Thermocatalytic formation of methane



Schematic view of the formation (modified after Froelich et al. 1979) and the subsequent pathways of methane in the sediment (modified after Judd 2004). Crystallographic image of gas hydrates after Bohrmann and Torres (2006)

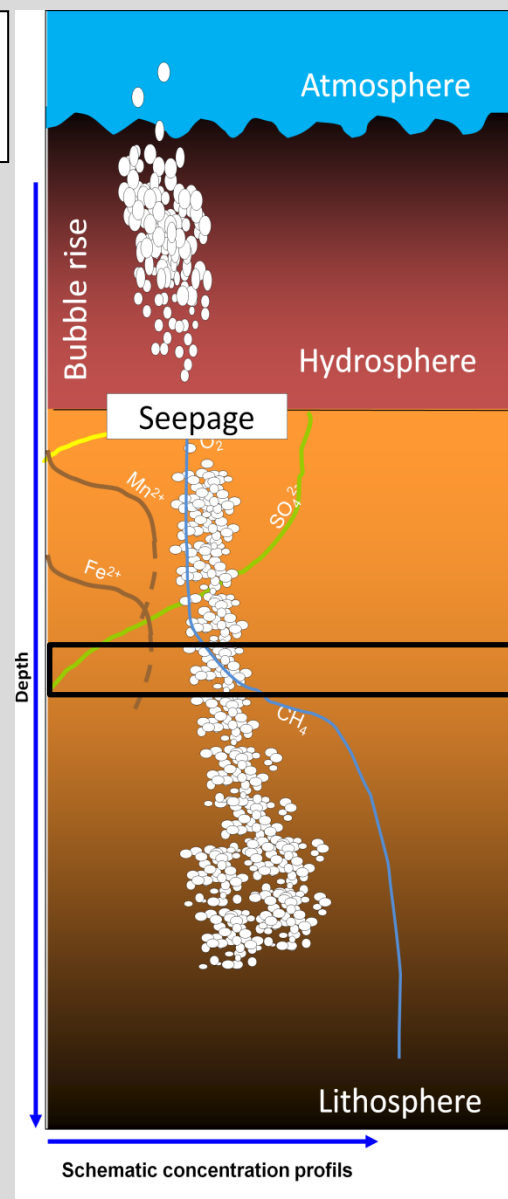
Storage and migration of methane



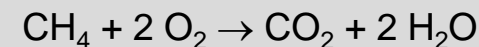
Schematic view of the formation (modified after Froelich et al. 1979) and the subsequent pathways of methane in the sediment (modified after Judd 2004). Crystallographic image of gas hydrates after Bohrmann and Torres (2006)

Utilization of methane in the sediment

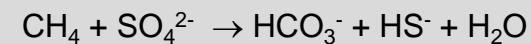
Only if the rate of methane production in relation of migration exceeds the rate of microbial utilization, seepage into the water column occurs.



Aerobic oxidation



Anaerobic oxidation of methane (AOM)



(Boetius et al. 2000)

Sulfate / Methane Transition Zone (SMTZ)

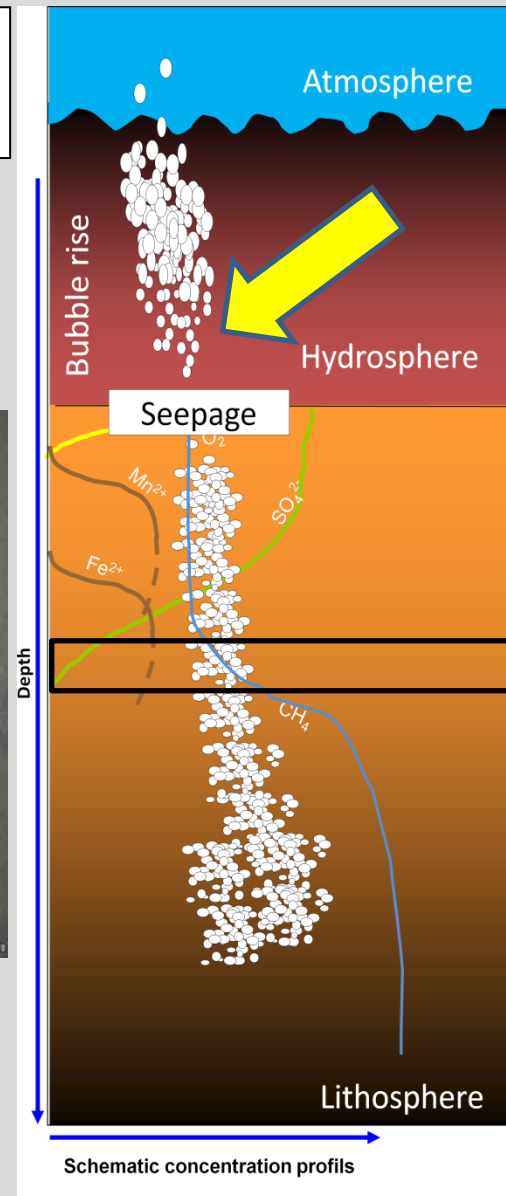
Free methane gas

Schematic view of the formation (modified after Froelich et al. 1979) and the subsequent pathways of methane in the sediment (modified after Judd 2004).

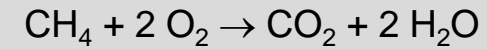
Utilization of methane in the sediment



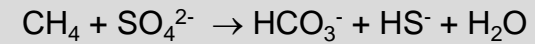
Heincke 362



Aerobic oxidation



Anaerobic oxidation of methane (AOM)



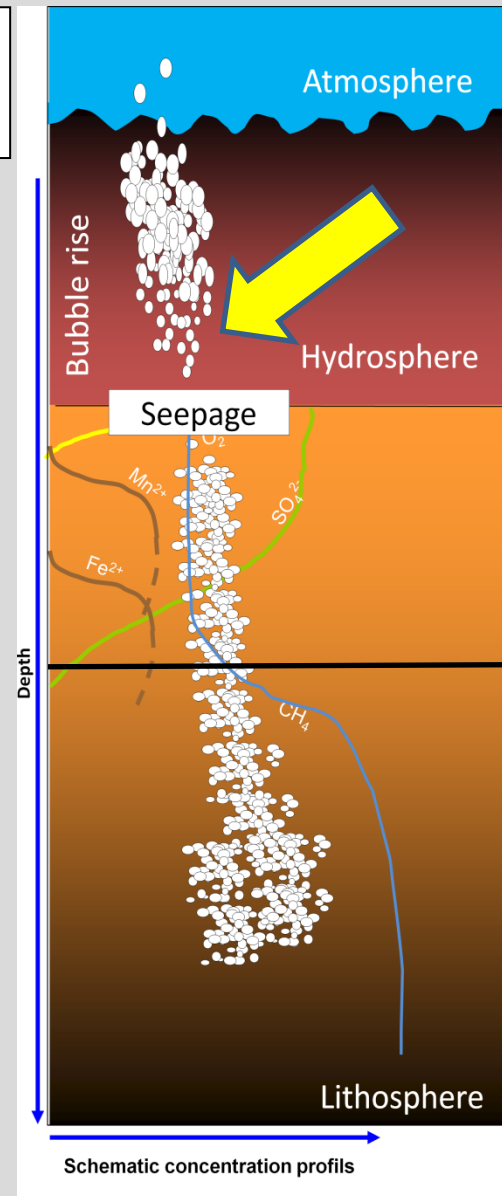
(Boetius et al. 2000)

Sulfate / Methane Transition Zone (SMTZ)

Free methane gas

Schematic view of the formation (modified after Froelich et al. 1979) and the subsequent pathways of methane in the sediment (modified after Judd 2004).

Pathways of methane in the water column



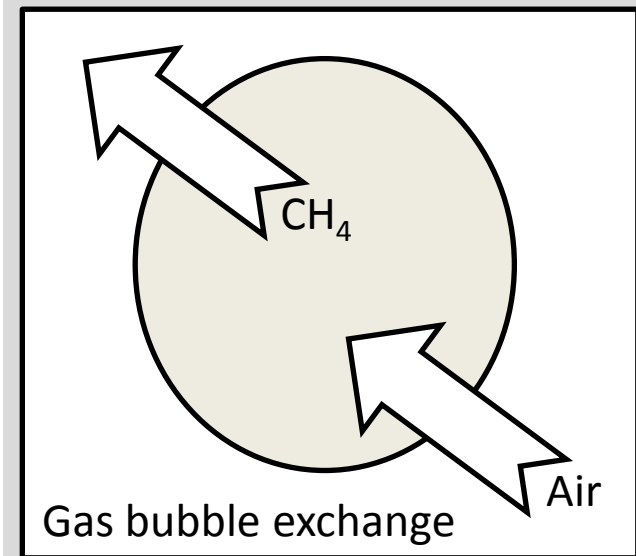
Air/Sea exchange

Vertical or horizontal transport of dissolved methane

Dilution

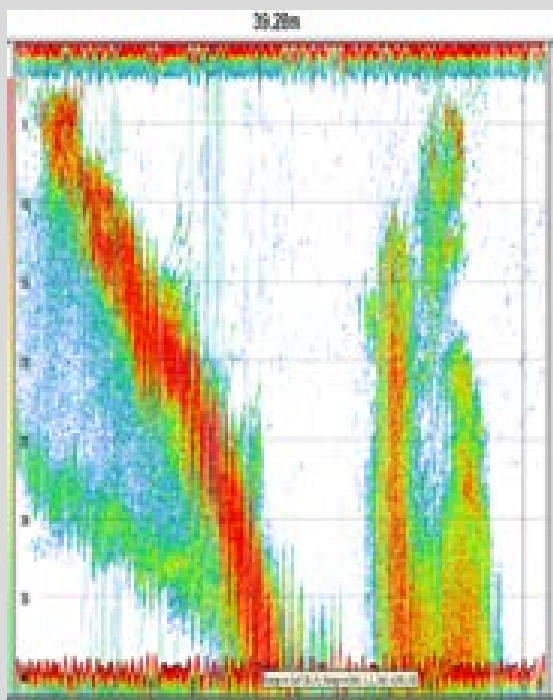
Microbial oxidation

Dissolution of methane from gas bubbles
(Epstein and Plesset 1950; Leifer and Patro 2002;
McGinnis et al. 2006)

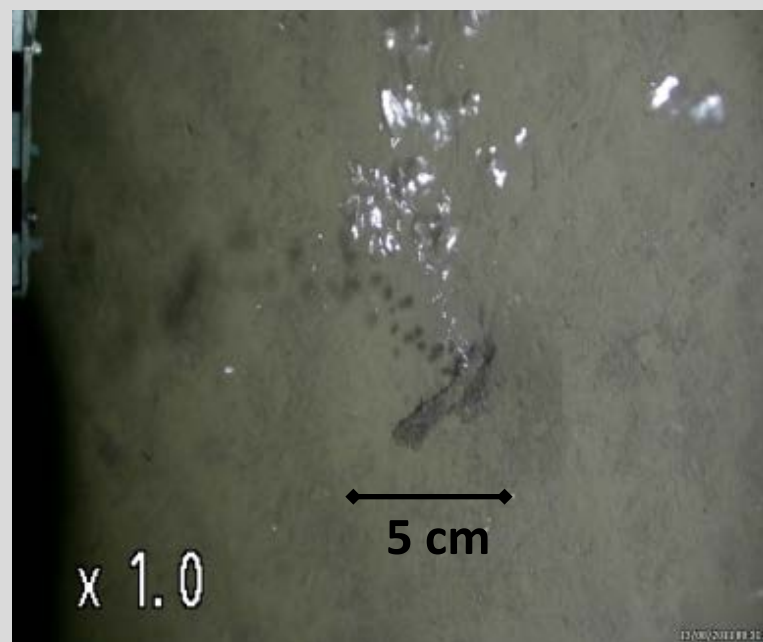


Schematic view of the formation (modified after Froelich et al. 1979) and the subsequent pathways of methane in the sediment (modified after Judd 2004).

HOW TO INVESTIGATE THE WATER COLUMN ABOVE GAS SEEPAGE?



Hydroacoustic “image” of gas bubble plumes in the water column by Simrad EK60.



Gas release in the North Sea via video observation

GAS ANALYSIS: STATE OF THE ART

Water column sampling

Phase separation:
gas phase from aqueous phaseGas analysis by gas
chromatography

(Lammers and Suess 1994)

Need for new methods

Problems:

- time consuming
- coarse spatial and temporal resolution

REQUIREMENTS FOR IN SITU SENSORS:

- Robustness for the use in harsh environment
- The energy consumption needs to be low to allow long term measurements
- Sampling rates should be high and respond times correspondingly short for high temporal and spatial resolution
- Maintenance of the analyzer should be easy and short in time
- A low detection limit for trace gases.

INSPECTR200-200 FOR IN SITU, ONLINE, REAL TIME AND SIMULTANEOUS MEASUREMENTS:



(Short et al. 2001)



Robustness for the use in harsh environment



The energy consumption needs to be low to allow long term measurements



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Maintenance of the analyzer should be easy and short in time

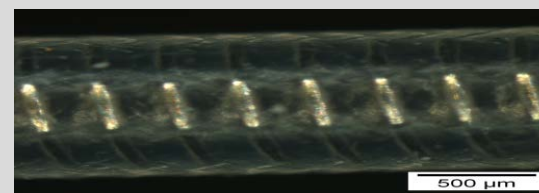


A low detection limit for trace gases.

IN SITU MASS SPECTROMETER MODE OF OPERATION



70 times magnification



320 times magnification

Water vapor

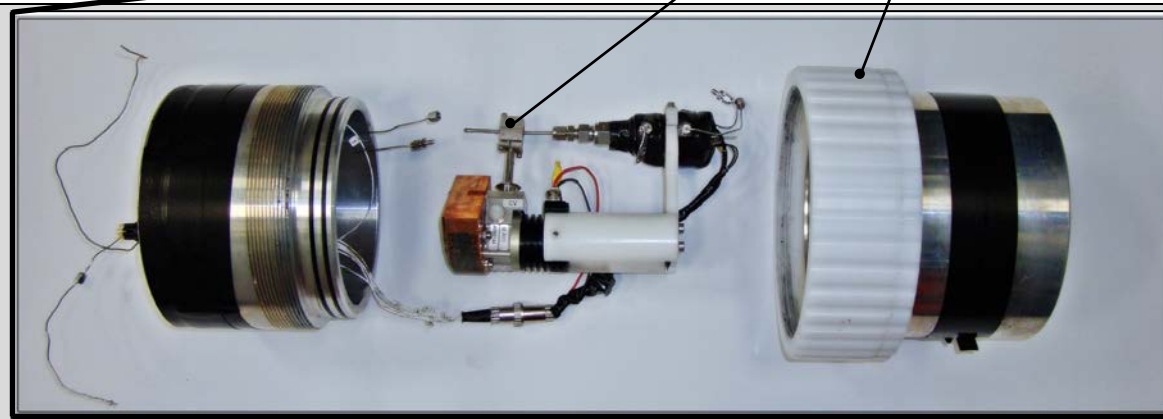
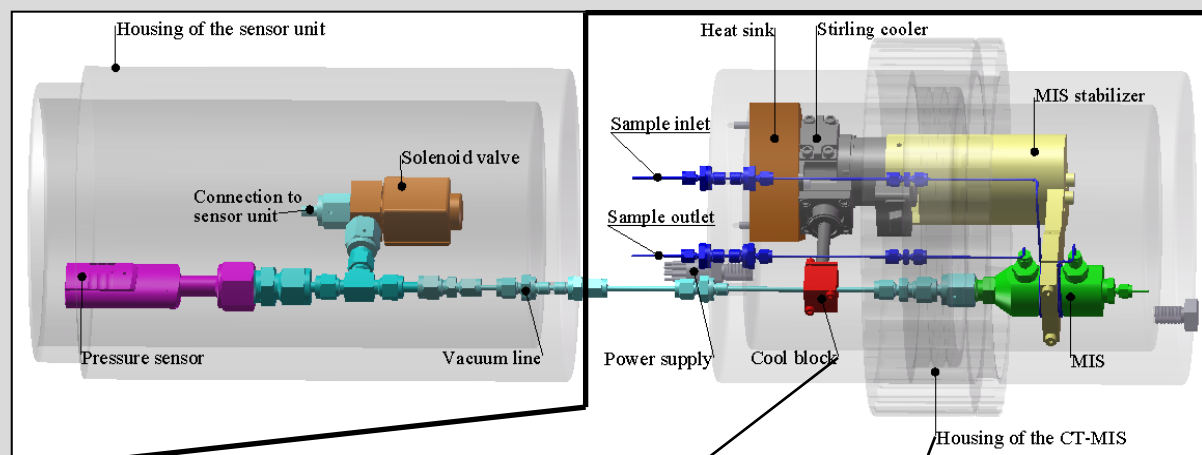
is the main gas that permeates through this membrane?

- Downgrades the detection limit
- Affects on the ionization efficiency
- Could cause condensation in the analytical line
- Downgrades the life time of the filament
- Indicate a high pressure in the analytical line

IMPLEMENTATION OF A CRYOTRAP



Micro Stirling Cooler, Ricor K508



(Gentz and Schlüter 2012)

Specifications:

Length: 290mm

Max depth: 200mm

Weight: 5.1 kg

Cooling area: 20mm

Outer diameter: 190mm

Inner diameter: 180mm

Material: Aluminum

IMPLEMENTATION OF A CRYOTRAP

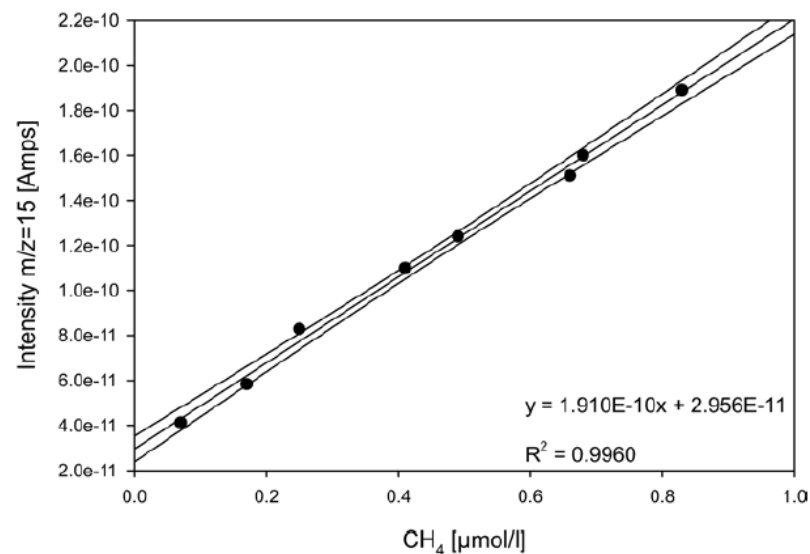
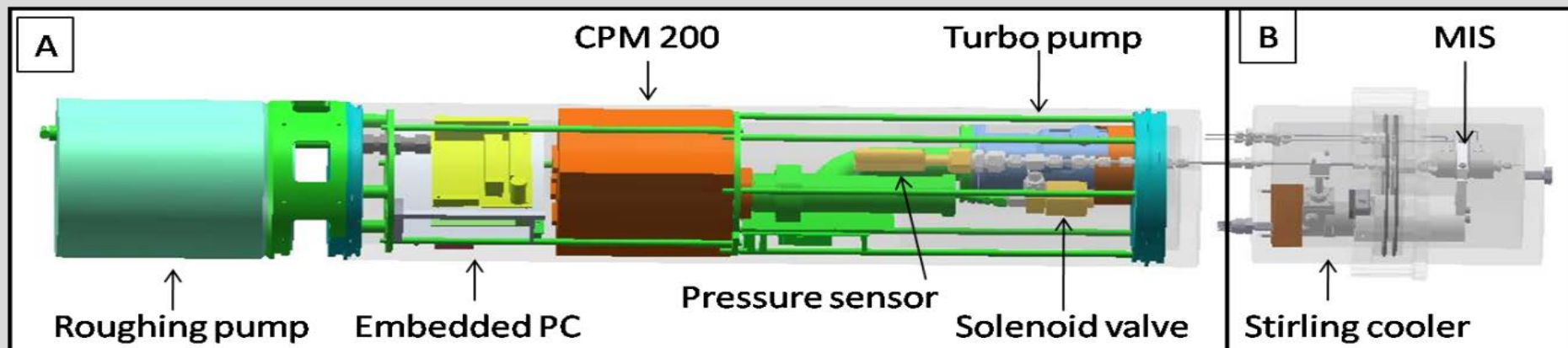
Cooling of the capillary between sample inlet and sensor unit up to -90 °C

- Water vapour is reduced up to 98 % of initial
- Reduce the internal pressure significantly
- A higher ionization efficiency is observed

→ Results in an optimized detection limit

- Expand the lifetime of the analyser
- Secure the analyser for inflowing water

OPTIMIZED AND REDESIGNED INSPECTR200-200

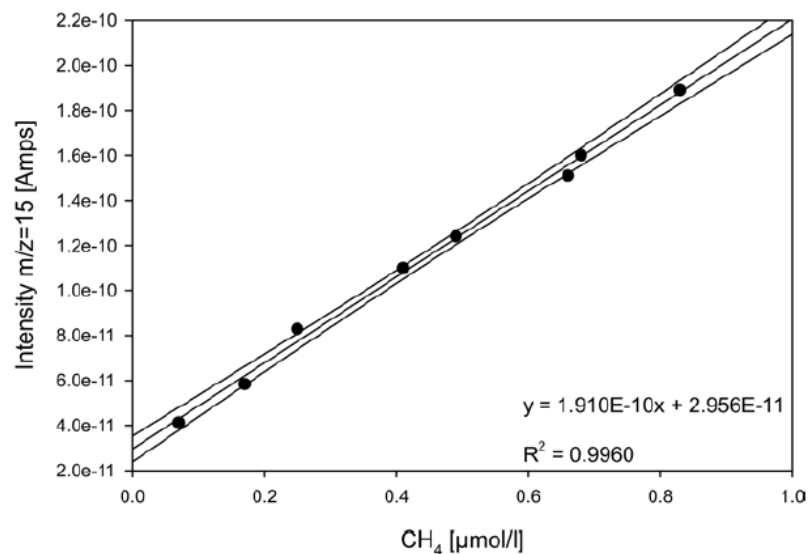
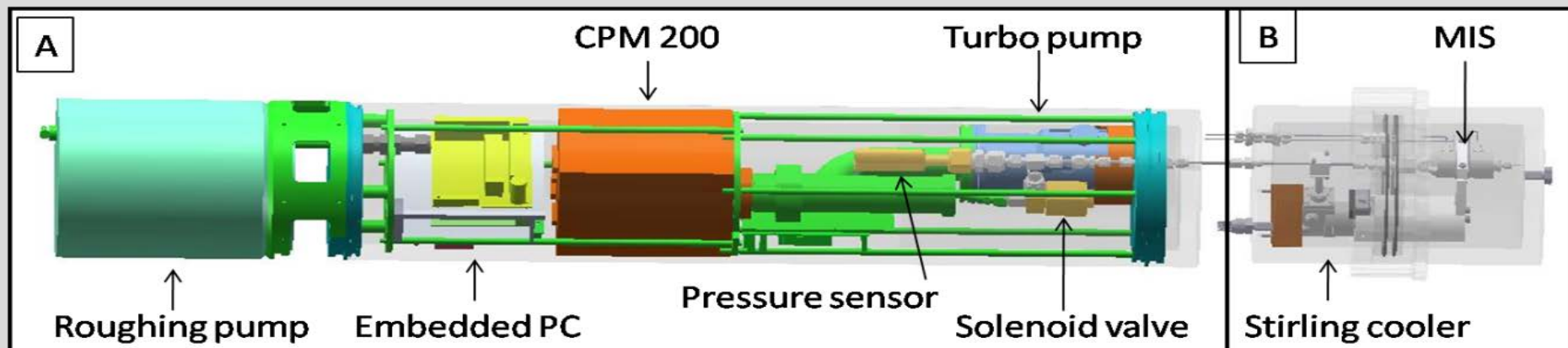


Calibration of the optimized Inspectr200-200

New detection limit of the optimized Inspectr200-200:

~16 nmol L⁻¹

OPTIMIZED AND REDESIGNED INSPECTR200-200



Calibration of the optimized Inspectr200-200

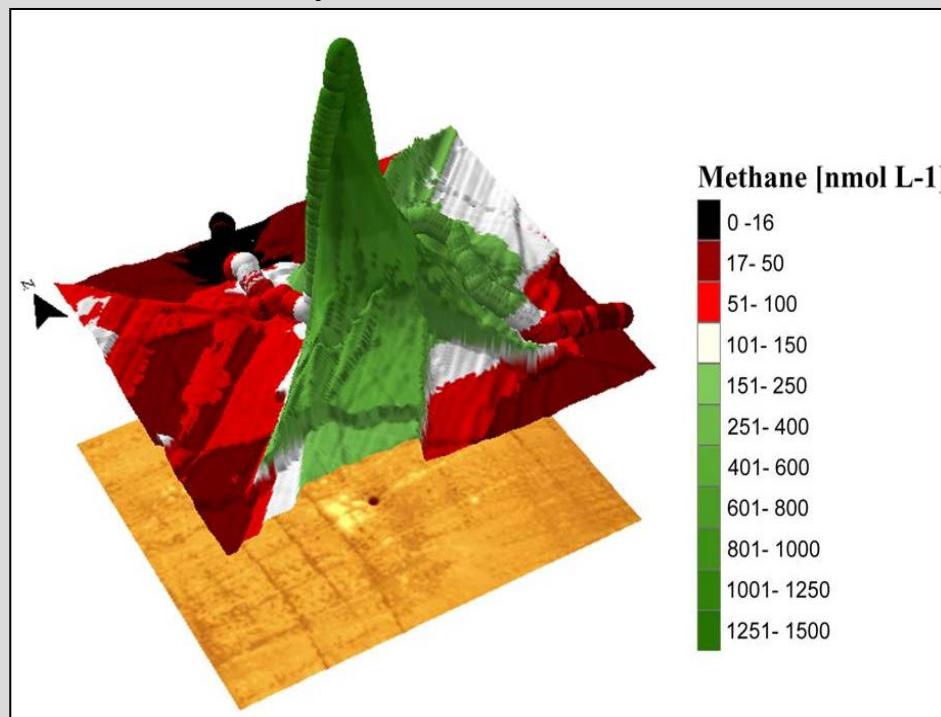
**New detection limit of the
optimized Inspectr200-200:**

~16 nmol L⁻¹

Low enough???

IN SITU MASS SPECTROMETER FOR FIELD APPLICATIONS

Gas seep in the North Sea

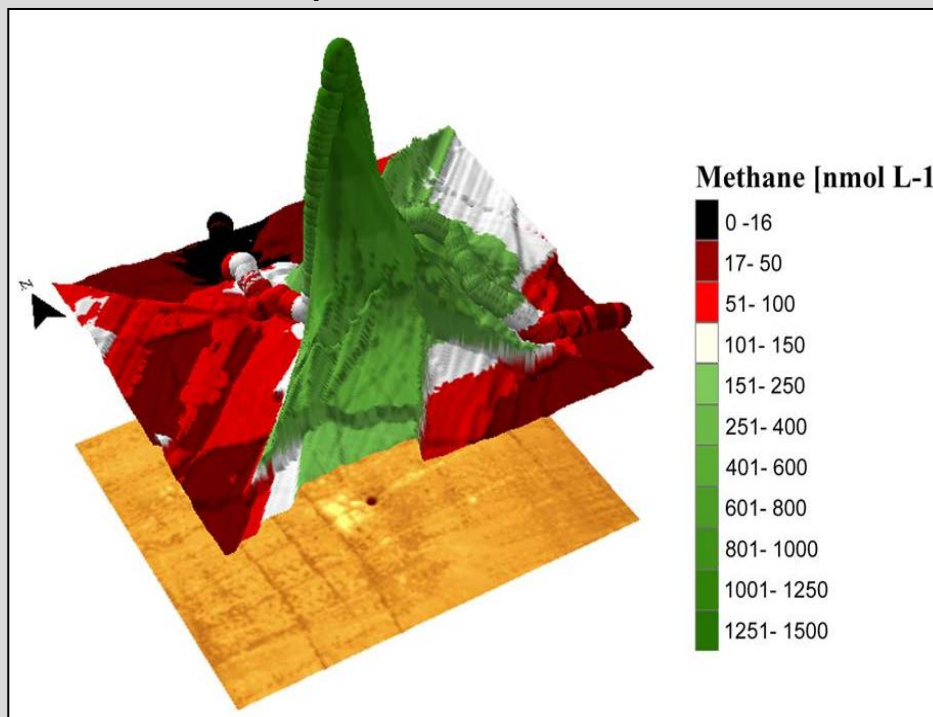


Concentration [nmol L ⁻¹]	Area [%]
< 16	3.6
16 - 100	48.3
> 100	48.1

IN SITU MASS SPECTROMETER FOR FIELD APPLICATIONS

Gas seep in the North Sea

without
cryotrap:
48.1 %

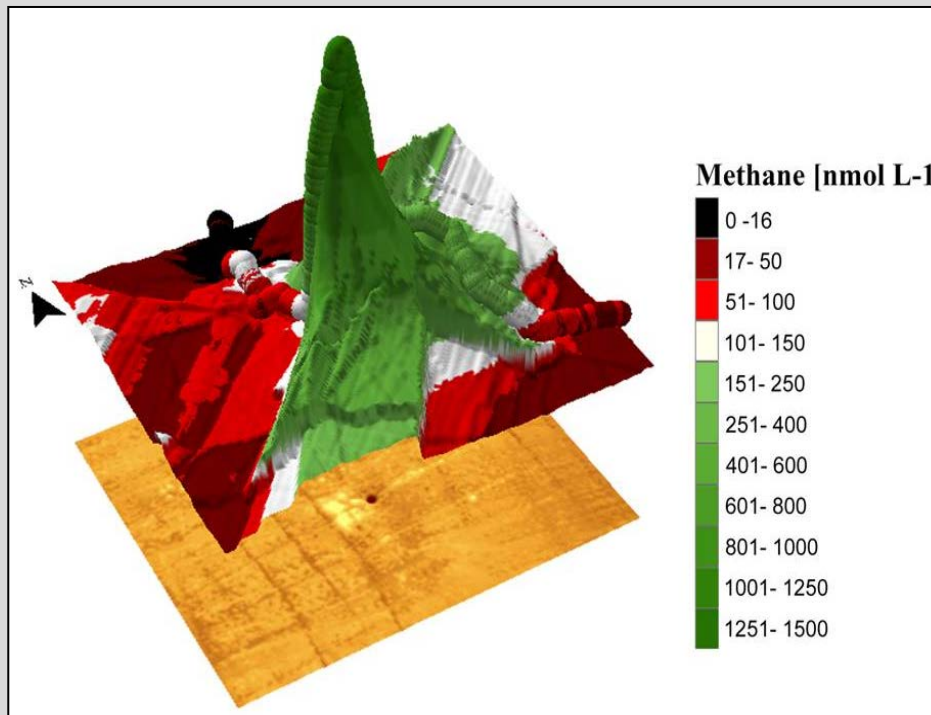


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IN SITU MASS SPECTROMETER FOR FIELD APPLICATIONS

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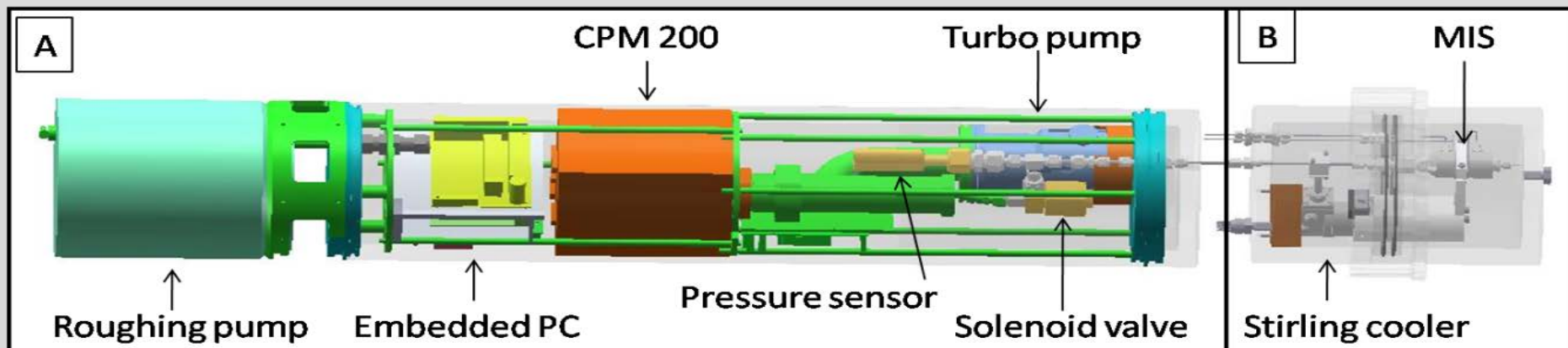


with
cryotrap:
96.4 %



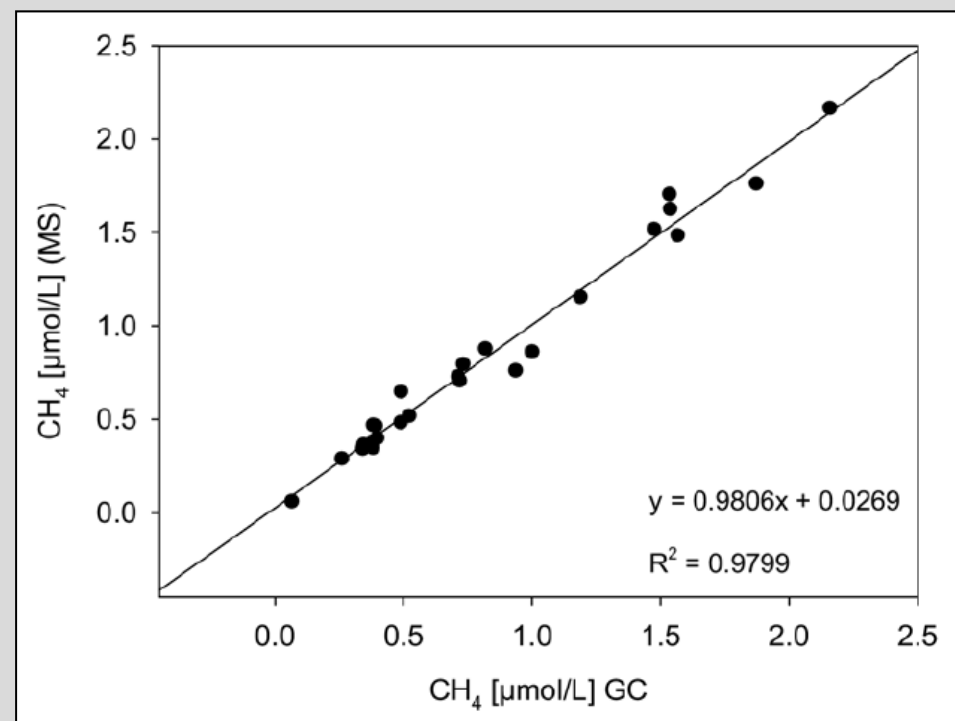
Concentration [nmol L ⁻¹]	Area [%]
< 16	3.6
16 - 100	48.3
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COMPARISON OF THE INSPECTR200-200 VS. CONVENTIONAL TECHNIQUES



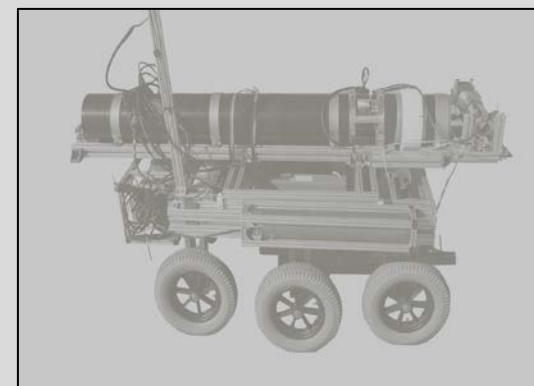
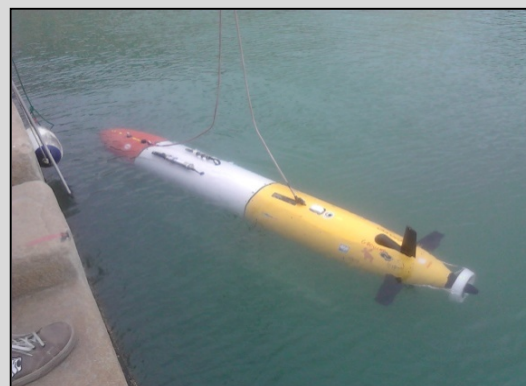
- Both methods are comparable
- No sampling preparation
- Simultaneous measurement of the dissolved gases
- No artefacts during sampling
- Up to 750 times higher sampling frequency

→ Higher temporal and spatial resolution



Inspectr200-200 vs. GC

FS Heincke

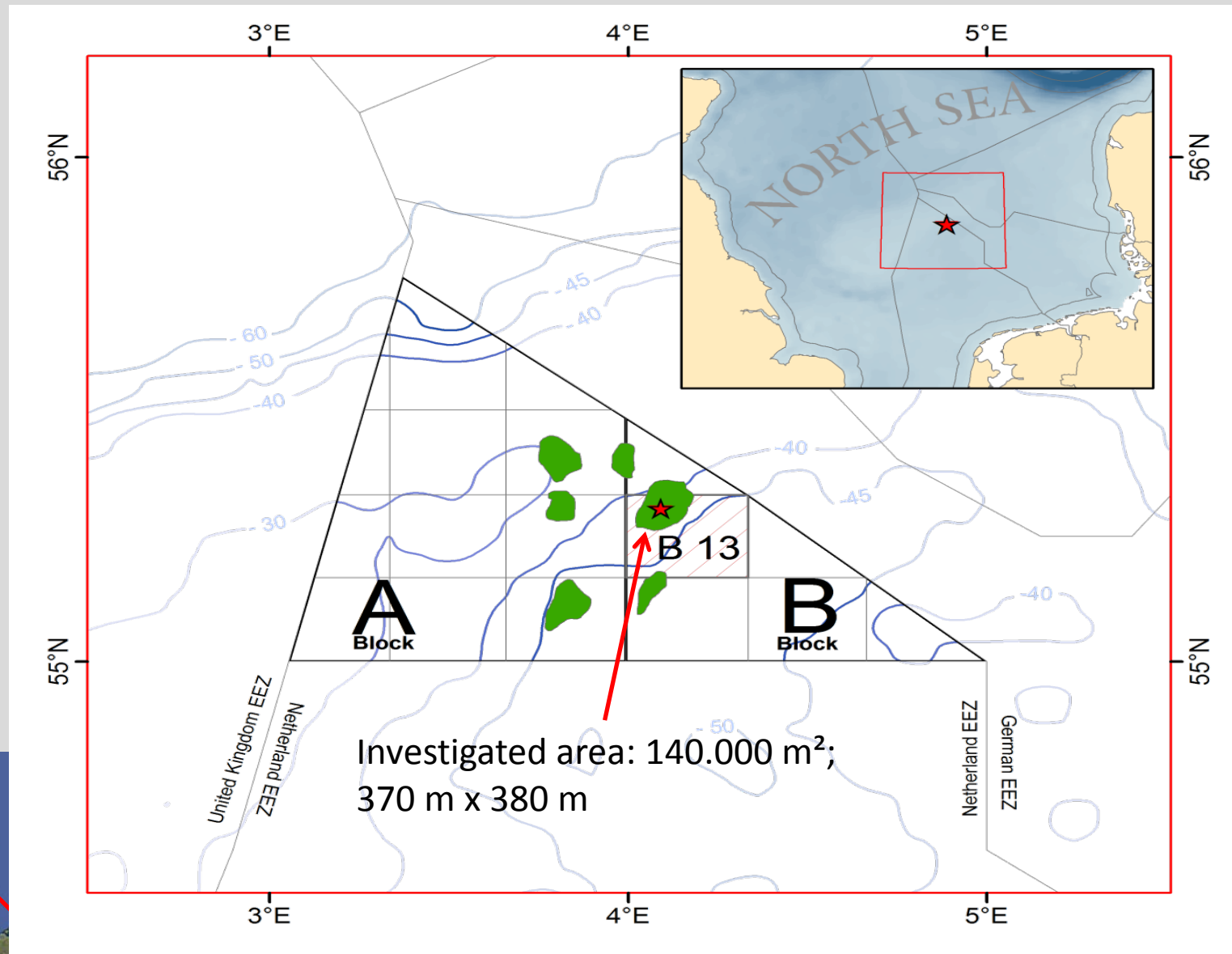


Deployments of the in situ mass spectrometer

APPLICATION OF THE IN SITU MASS SPECTROMETER IN HARSH ENVIRONMENTS

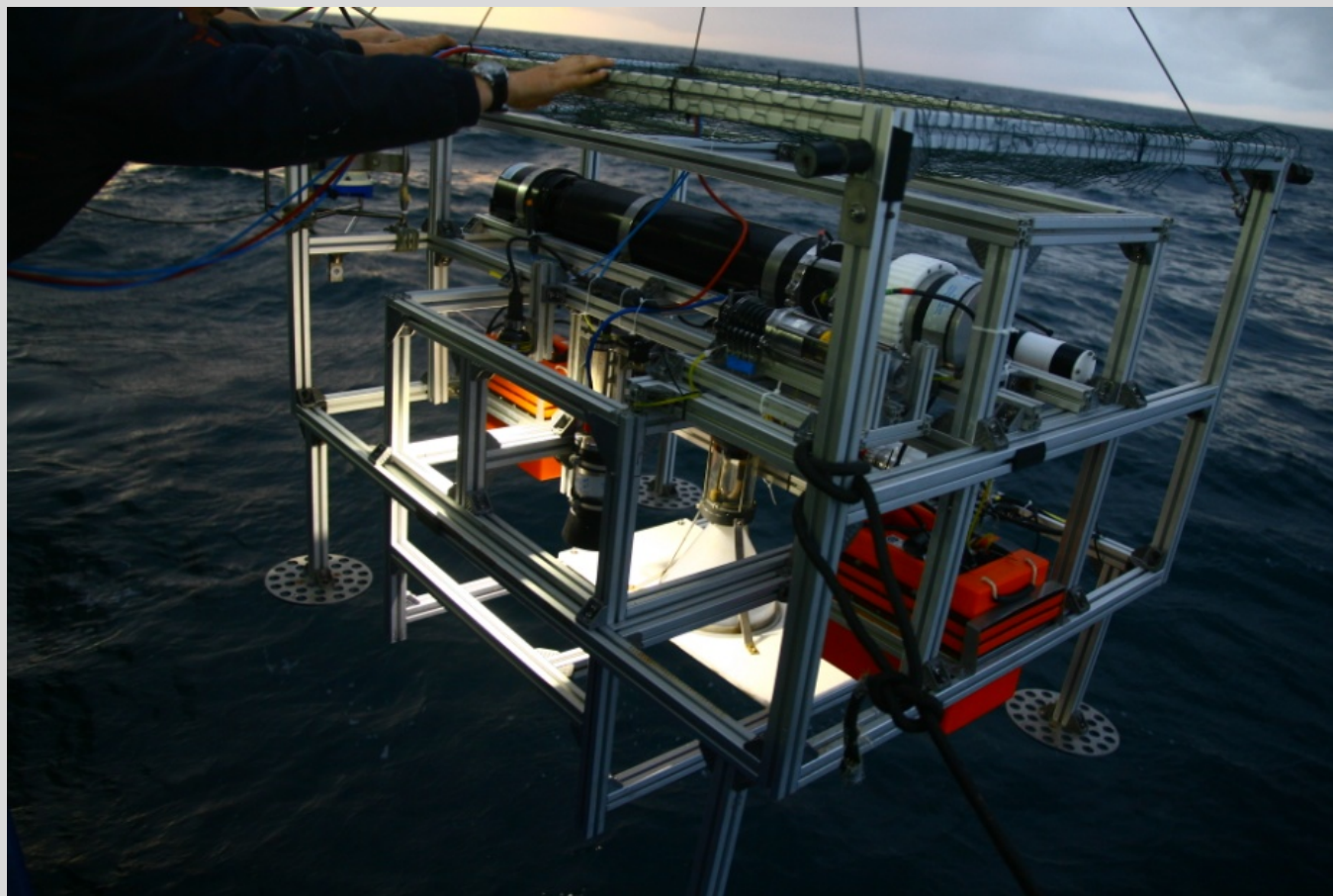


OBSERVATION OF A GAS SEEP AREA IN THE NORTH SEA



Modified after Schroot et al. 2005

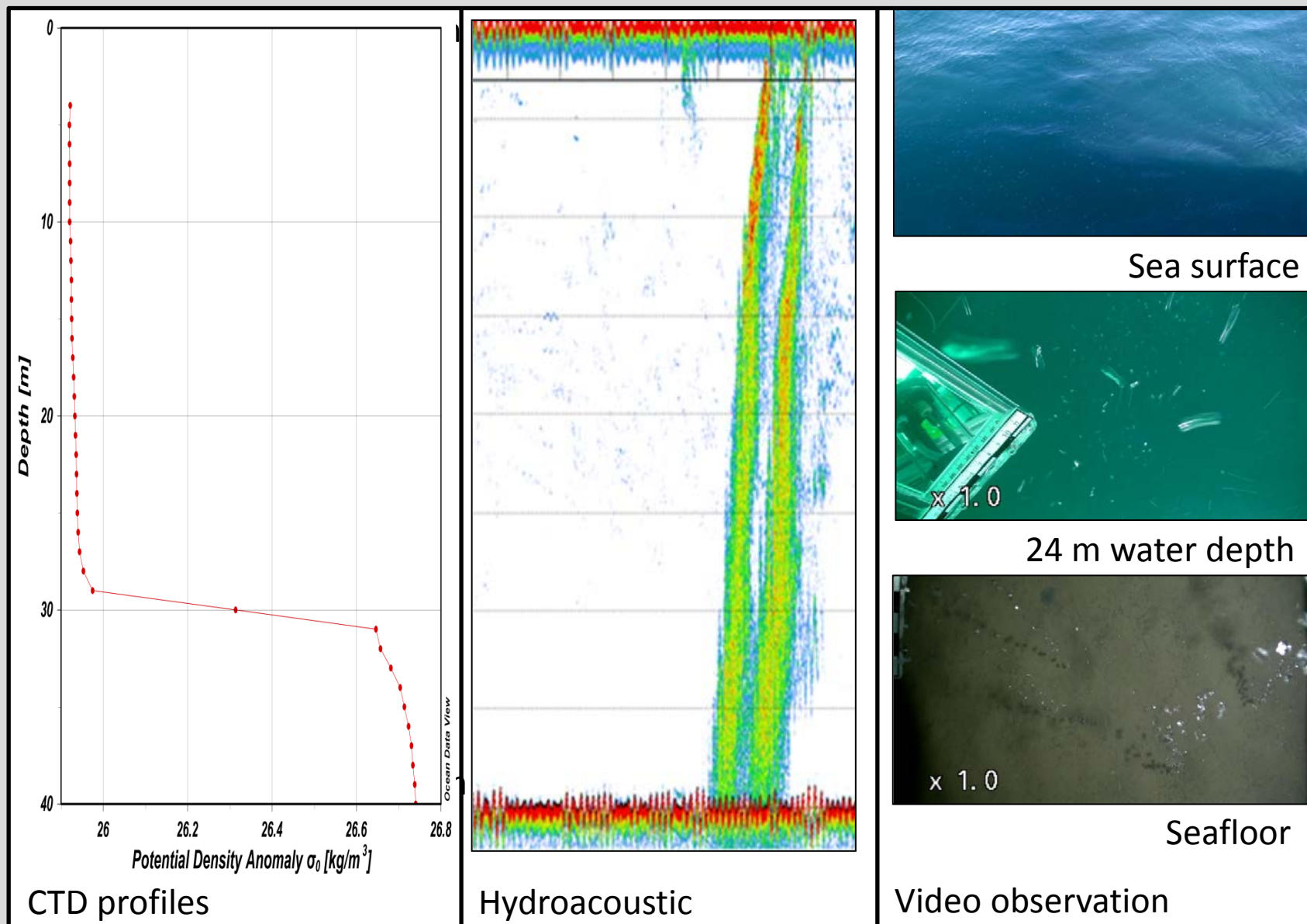
OBSERVATION OF A GAS SEEP AREA IN THE NORTH SEA



Under water gas analyser, sampler and observing system

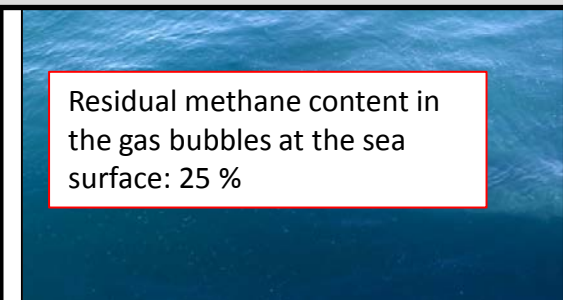
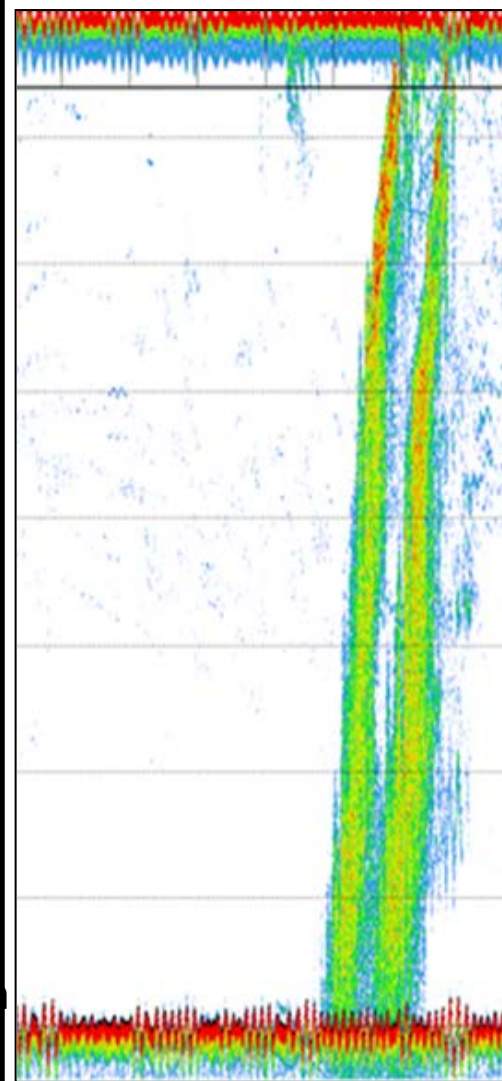
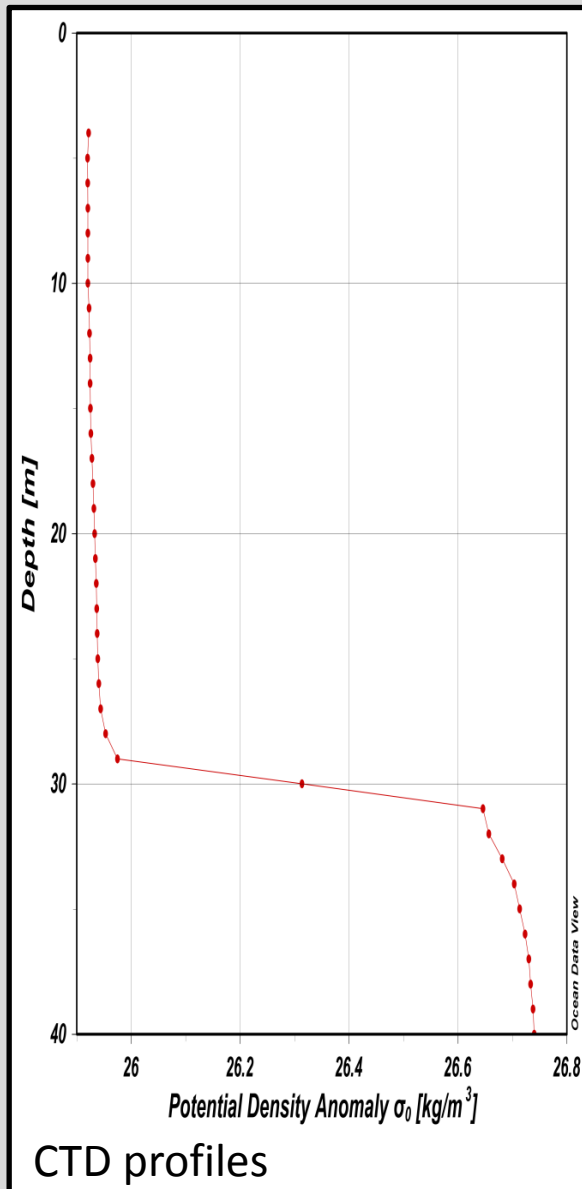
- Inspectr200-200; 11900 samples
- GC; discrete 154 samples
- Video observation; 12 h
- Hydroacoustic; 12 h
- Multibeam; 140000 m²
- CTD 14; vertical profiles
- Bubble sampler; 5 samples
- Multiple sediment corer; 5 cores

OBSERVATION OF A GAS SEEP AREA IN THE NORTH SEA



(Gentz et al. unpublished data)

OBSERVATION OF A GAS SEEP AREA IN THE NORTH SEA



Sea surface



24 m water depth

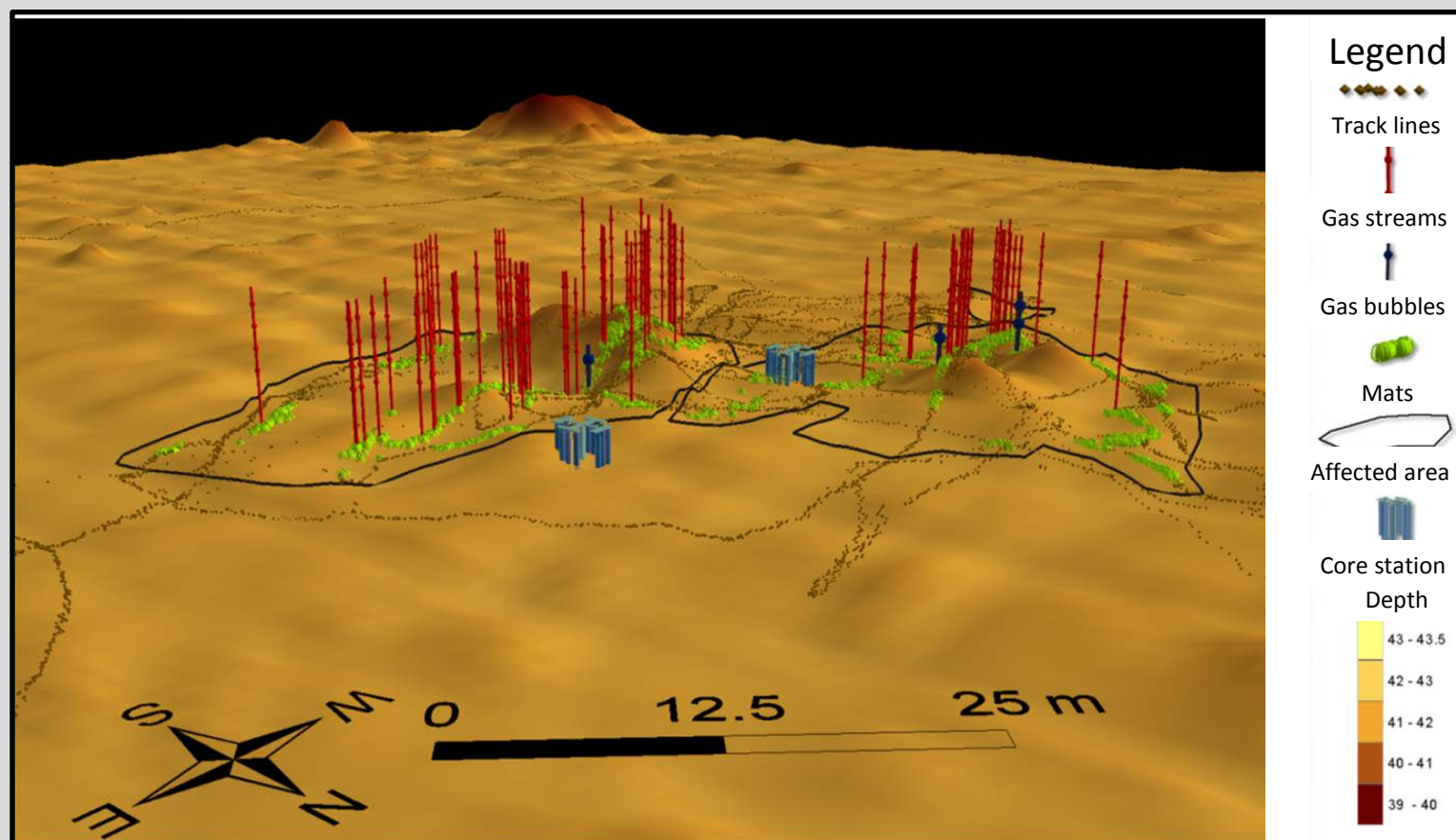


Seafloor

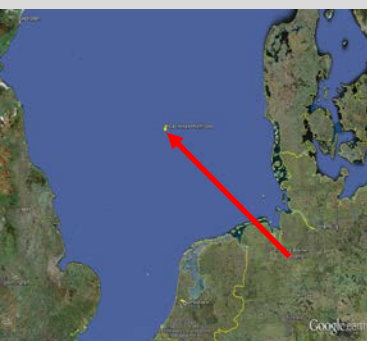
Video observation

(Gentz et al. unpublished data)

VIDEO OBSERVATION OF THE SEAFLOOR



(Gentz et al. unpublished data)

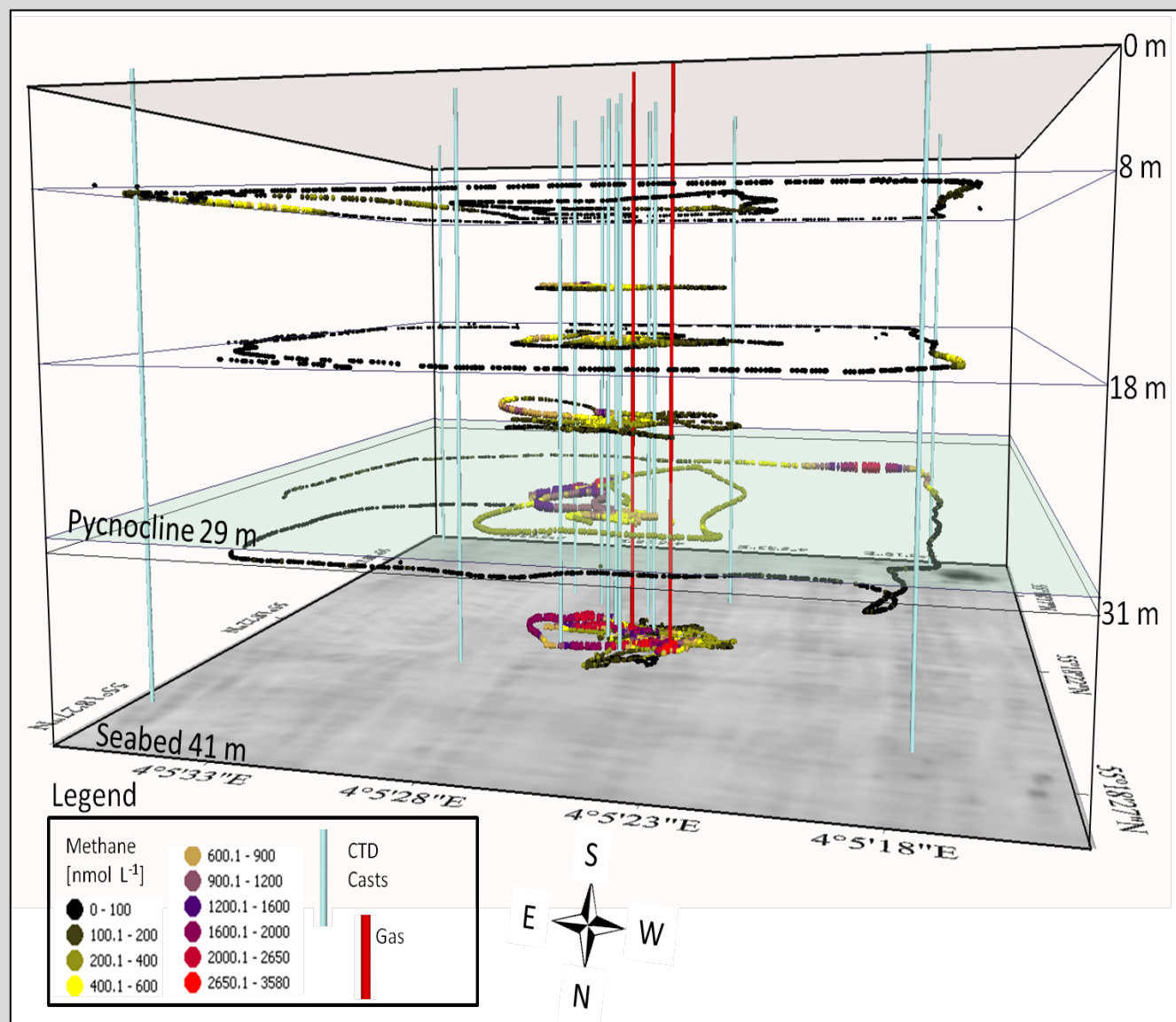
Affected area: $\sim 3800 \text{ m}^2$

Number of streams: 113

Bubble diameter: 4.5 to 16 mm (average 7 mm)

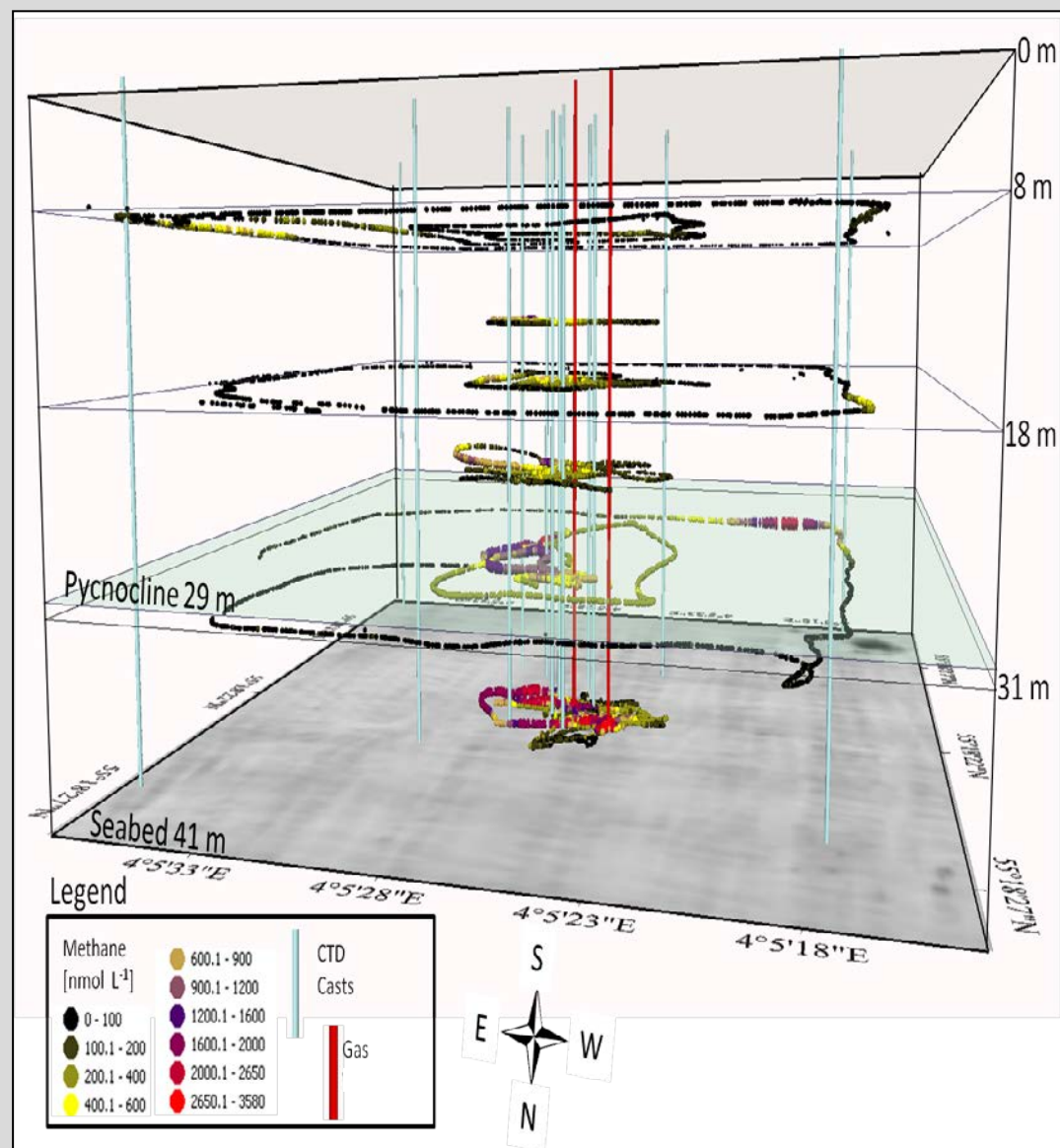
Release frequency: $0.3 - 40 \text{ bubbles s}^{-1}$ (average $23 \text{ bubbles s}^{-1}$)Methane flux: 28.27 L min^{-1} Methane release: $35.3 \pm 17.65 \text{ t CH}_4 \text{ yr}^{-1}$

DISSOLVED METHANE SAMPLING IN THE WATER COLUMN



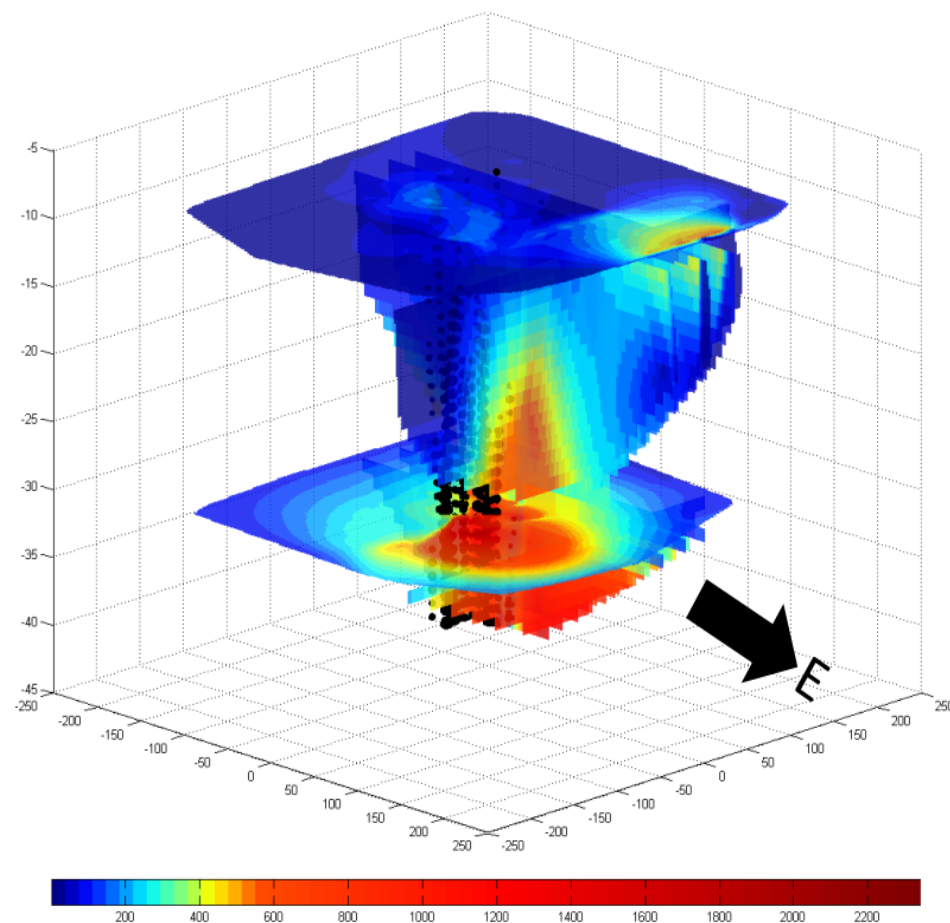
11900 samples in various depth in between 24 hours

DISSOLVED METHANE SAMPLING IN THE WATER COLUMN



- Discrete sampling: max 1.5 $\mu\text{mol L}^{-1}$
- In situ sampling: max 3.5 $\mu\text{mol L}^{-1}$
- A methane saturation of 23200 % was observed in 8 m water depth.
- The air sea exchange flux is calculated to $\sim 210 \pm 63 \mu\text{mol m}^{-2} \text{d}^{-1}$.

INTERPOLATION OF THE DISSOLVED METHANE ABOVE THE GAS SEEP



Entire interpolated inventory of methane ($6.410.000 \text{ m}^3$):

$\sim 0.6 \text{ mol CH}_4$

- $\sim 1.000.000 \text{ m}^3$ (15.6 %) contain concentrations higher than 200 nmol L^{-1}
- 40 % of initial methane is dissolved above the pycnocline.

MAIN RESULT NORTH SEA

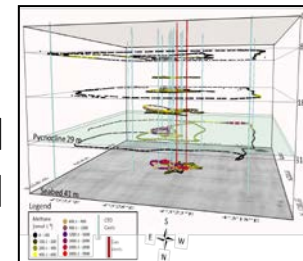
→ In total 65 % ($23 \pm 11.5 \text{ t CH}_4 \text{ y}^{-1}$) of the released methane potentially reach the atmosphere, which is high compared to the Spitsbergen continental margin or the Tommeliten area.

This is the first study of methane above a gas seep in high resolution.

CONCLUSIONS

This is the first study of methane above a gas seep in high resolution.

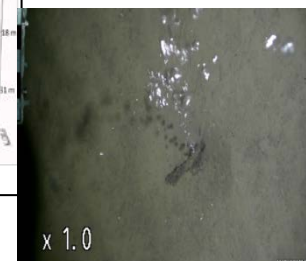
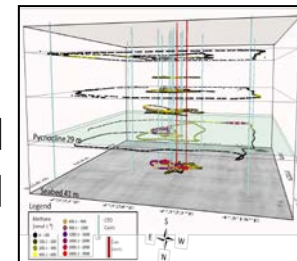
- The inventory calculation is more accurate than before and shows that conventional methods tend toward underestimation.



CONCLUSIONS

This is the first study of methane above a gas seep in high resolution.

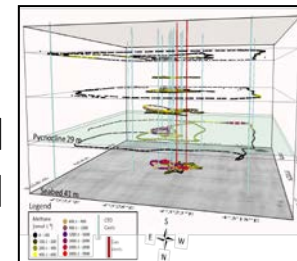
- The inventory calculation is more accurate than before and shows that conventional methods tend toward underestimation.
- The investigated study area in the North Sea contributes to the global atmospheric methane budget.



CONCLUSIONS

This is the first study of methane above a gas seep in high resolution.

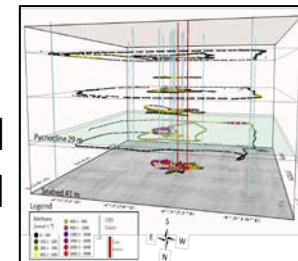
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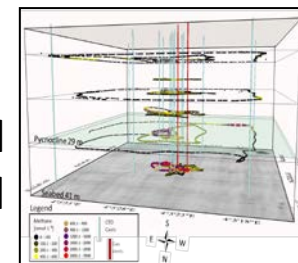


- The fate of methane as well as the contribution to the global atmospheric methane budget of each source depends on bubble size, the water depth, the water current and the water stratification.

	Spitsbergen	North Sea
Water depth [m]	245	40
Water stratification [m above seafloor]	25	10
Observed bubble rise [m above seafloor]	150	40
Estimated bubble diameter [mm]	< 5	7
Bubbles at seasurface	No	Yes
Direct methane transport	No	Yes
indirect transport	???	Yes
Methane to atmosphere [% from origin]	???	~ 60

This is the first study of methane above a gas seep in high resolution.

- The inventory calculation is more accurate than before and shows that conventional methods tend toward underestimation.
- The investigated study area in the North Sea contributes to the global atmospheric methane budget.



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Water depth [m]	245	40
Water stratification [m above seafloor]	25	10
Observed bubble rise [m above seafloor]	150	40
Estimated bubble diameter [mm]	< 5	7
Bubbles at seasurface	No	Yes
Direct methane transport	No	Yes
Indirect transport	???	Yes
Methane to atmosphere [% from origin]	???	~ 60

- The use of the improved in situ mass spectrometry is one step forward to understand the pathways and potential global relevance of these methane sources.**

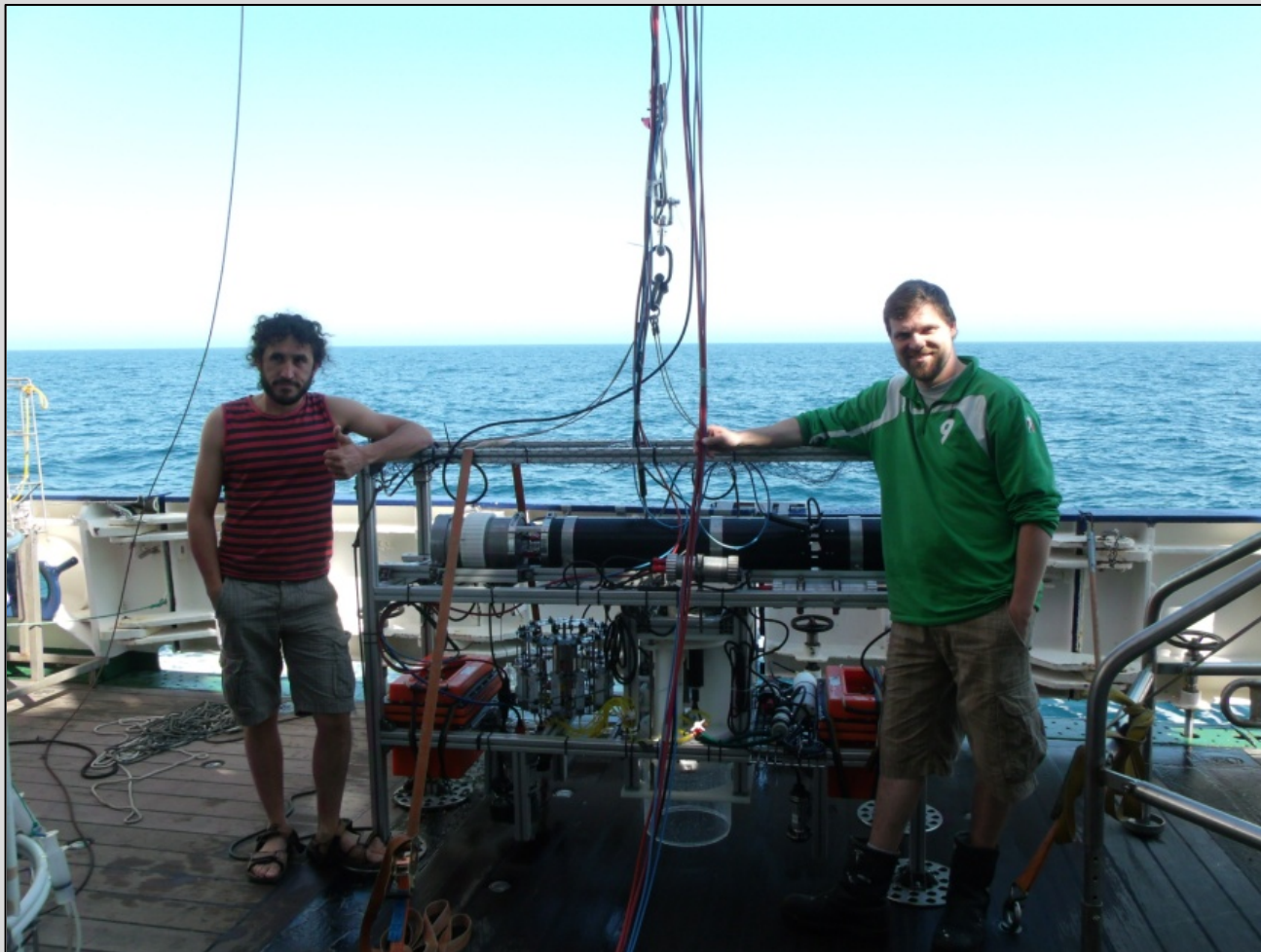


CURRENT AND FUTURE WORK

Implementation of the mass spec into an AUV

...





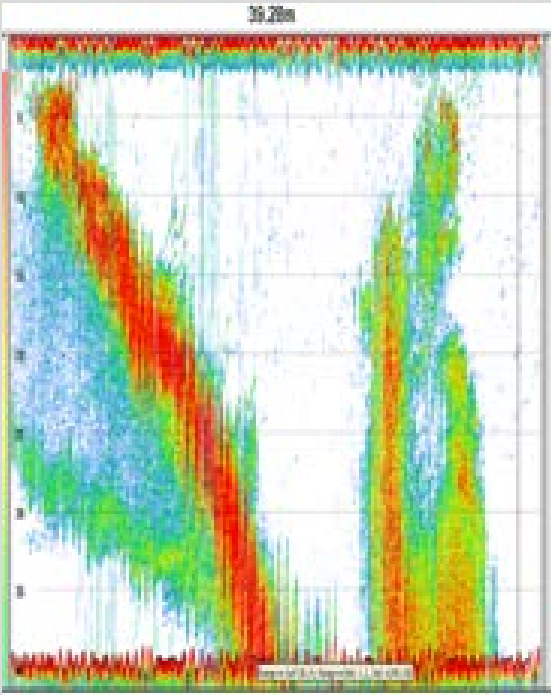
Thank you all for
developing new instruments as well as
your attention!

Backup

FUTURE WORK

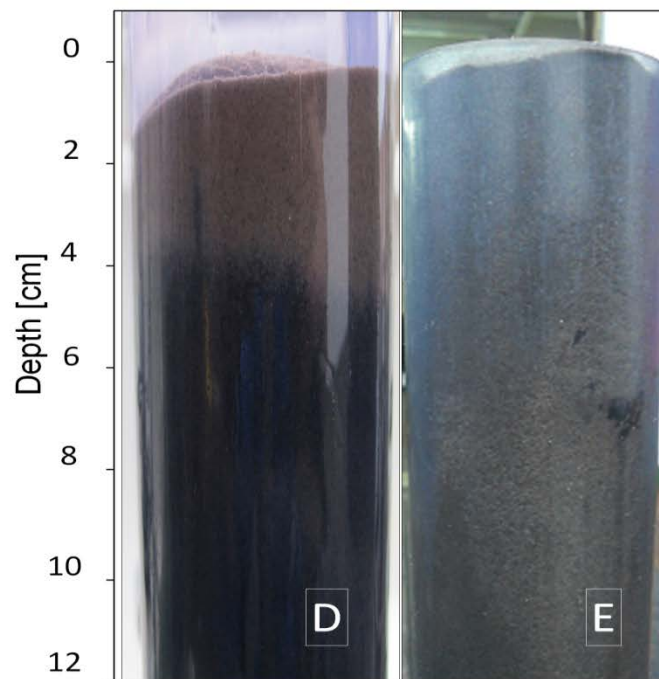
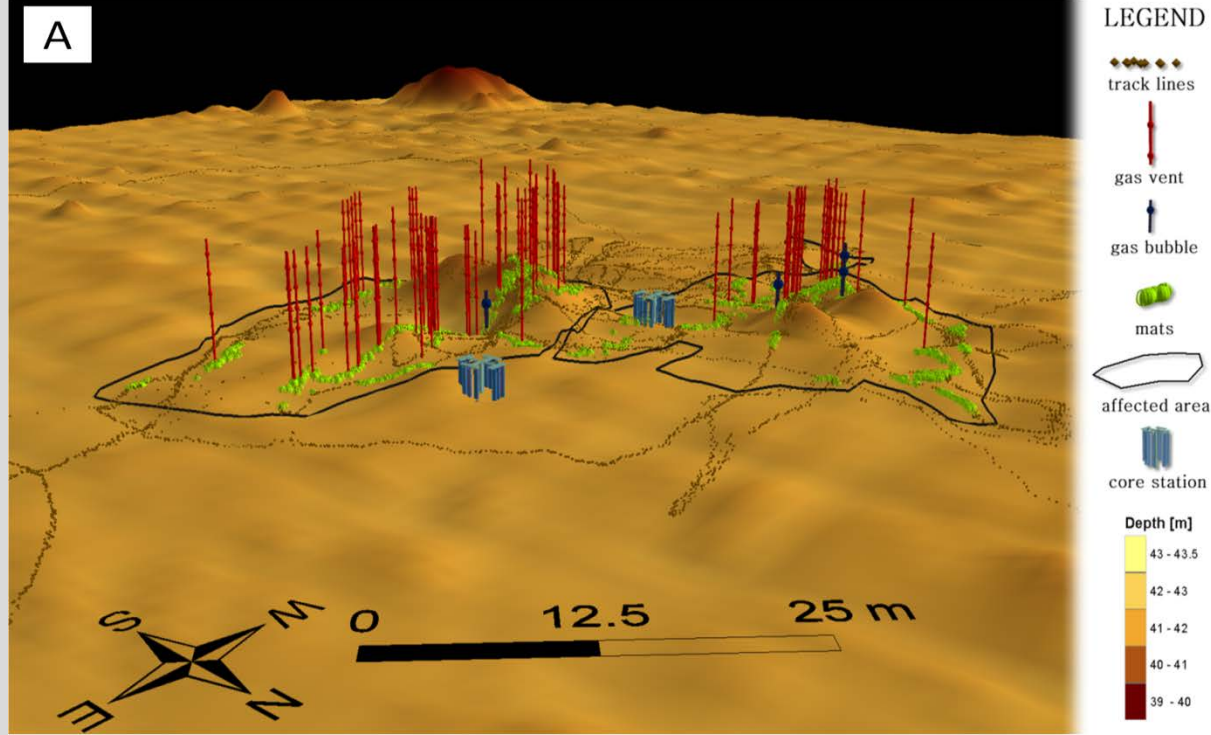


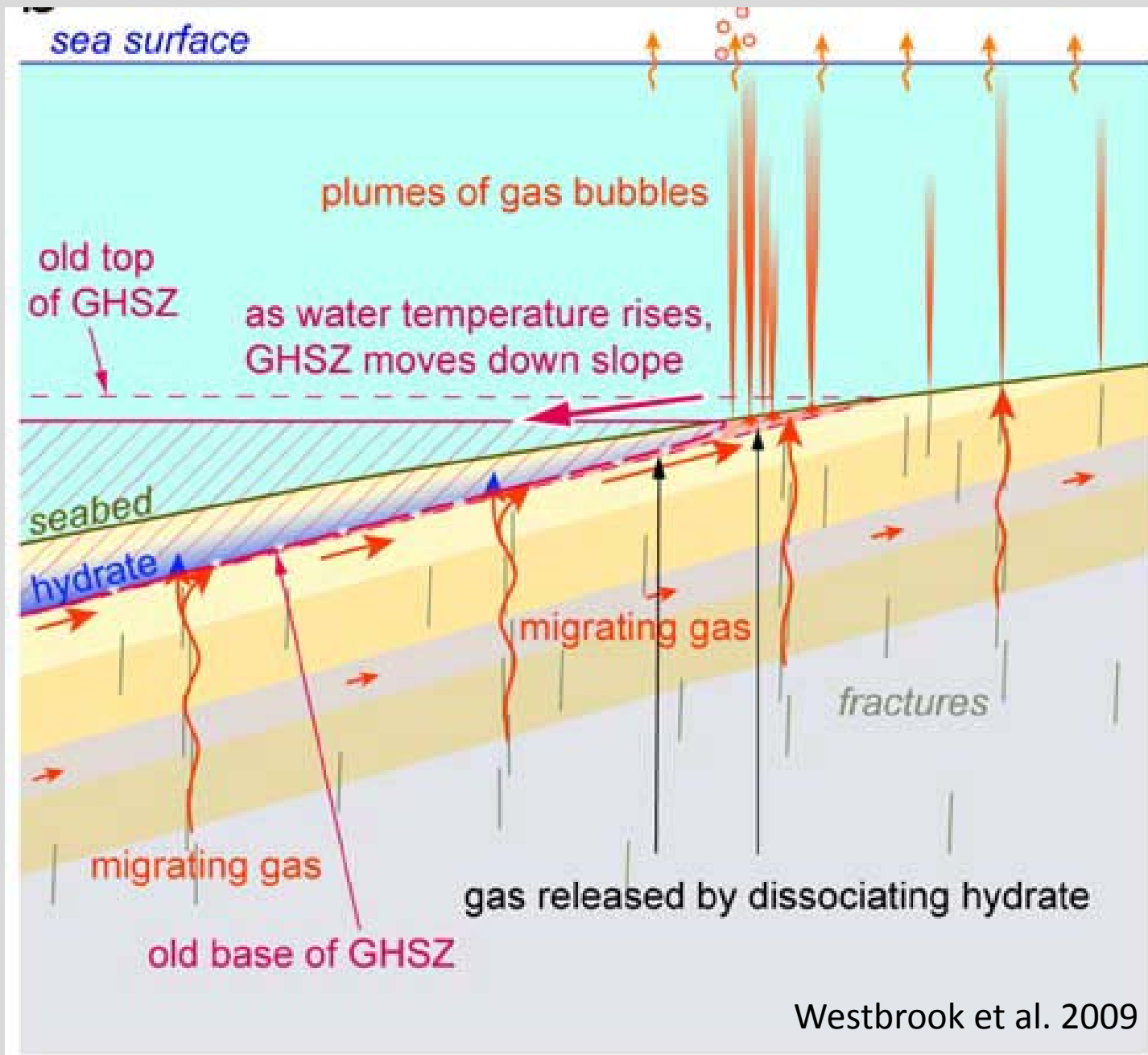
Implementation in new device holder



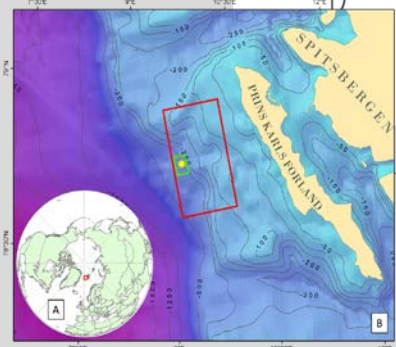
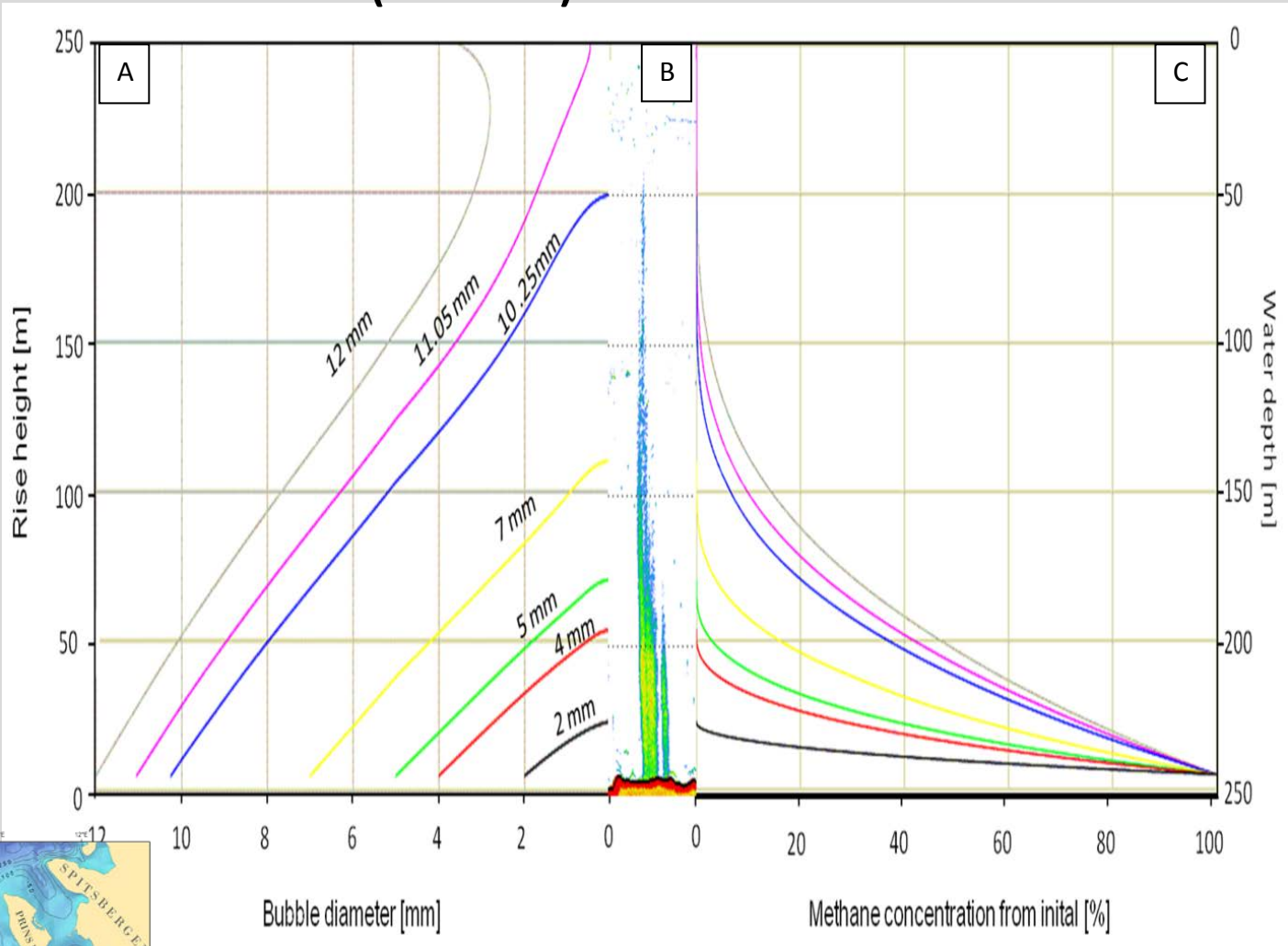
Benthic chamber measurements

Combining hydroacoustic with in situ mass spectrometry



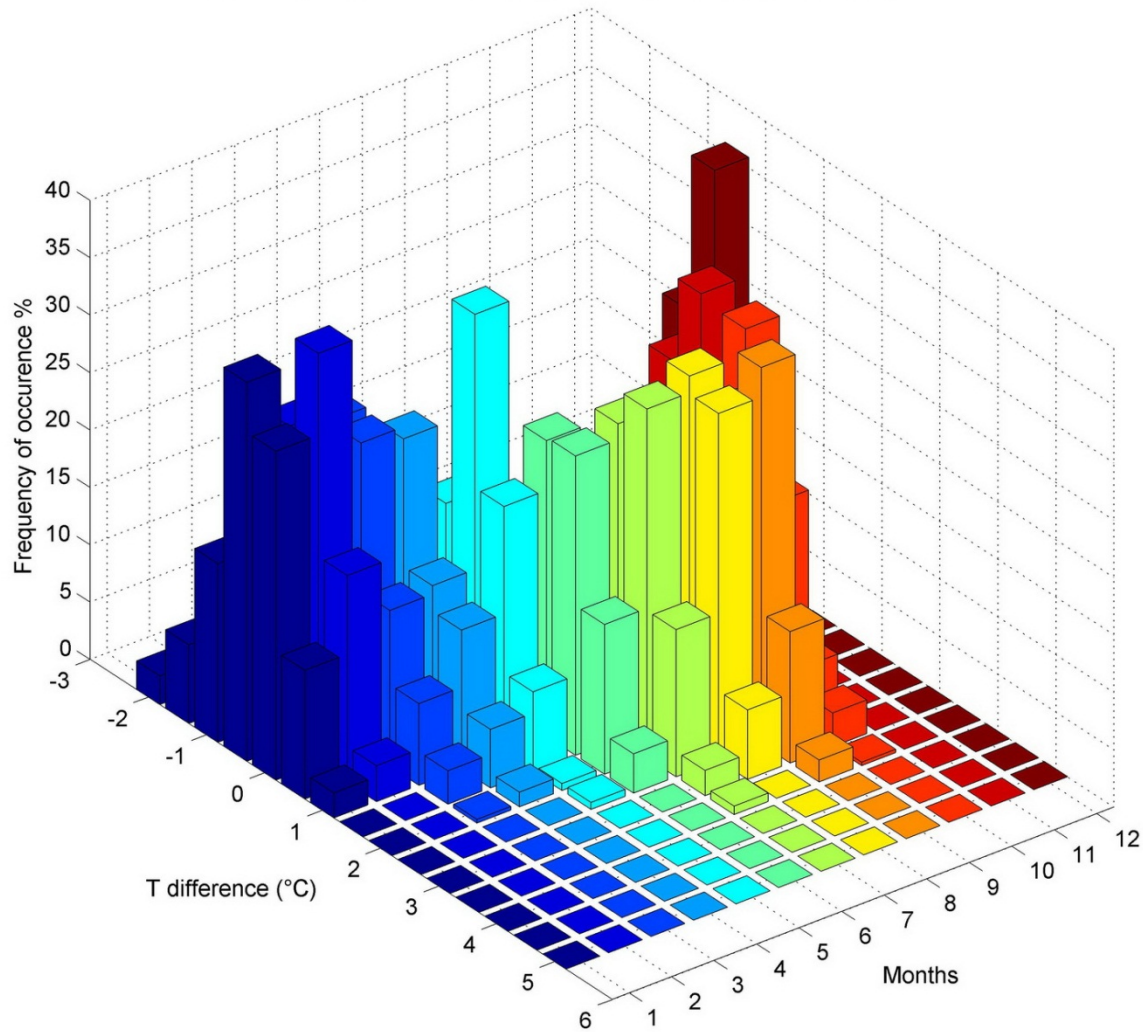


GAS BUBBLE DISSOLUTION MODEL (SiBU GUI):

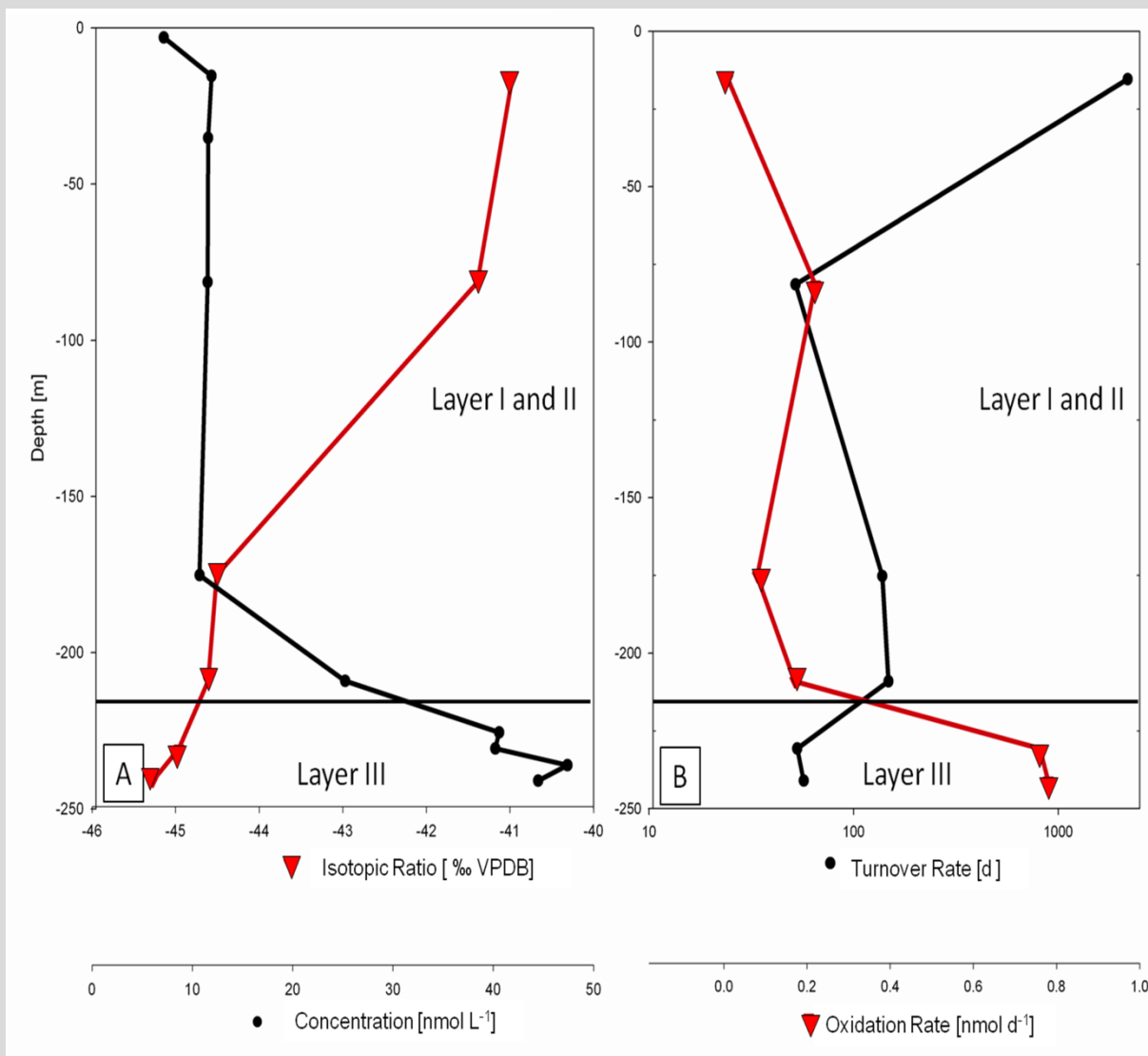


Decrease of the bubble diameter during the ascend from the seafloor for initial bubbles sizes of 2 mm to 12 mm (A) compared with the hydroacoustic image of the highest detected gas flare (B). Decrease of the initial CH₄ concentration in the bubbles during their rise in the water column (C). Data obtained by the model SiBU GUI (Greinert, J. and D. F. McGinnis 2009) personally optimized by Dan McGinnis

Monthly distributions of temperature differences between the subsurface (~50m) and near-bottom (~270m) layers at mooring F1 (78°50'N 8°40'E) in 1997-2010 based on daily averaged data

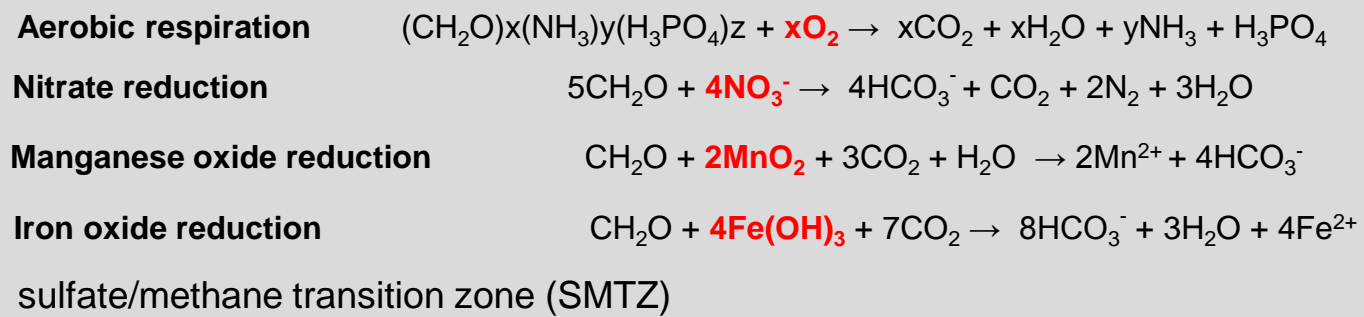
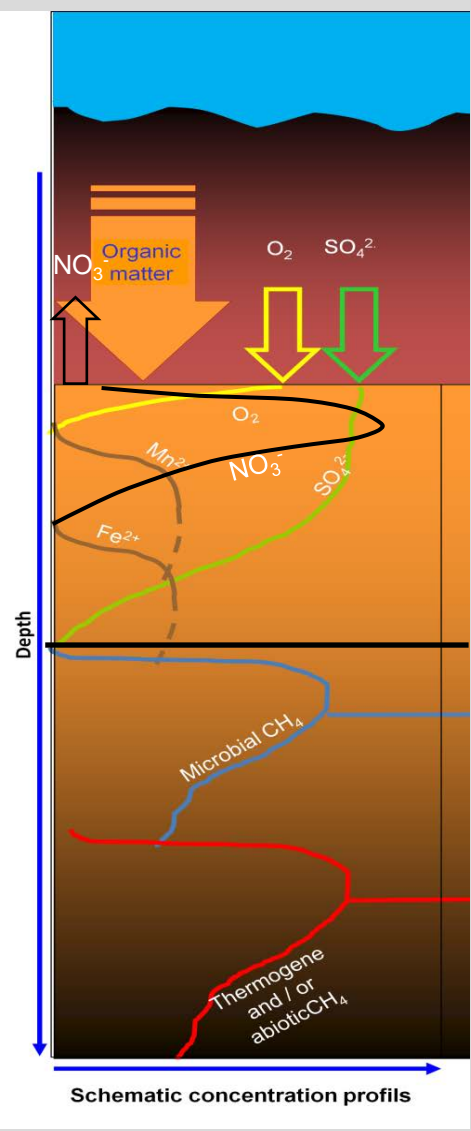


Personal communication Agnieszka Beszsynsky-Möller
28.26 km s-w direction

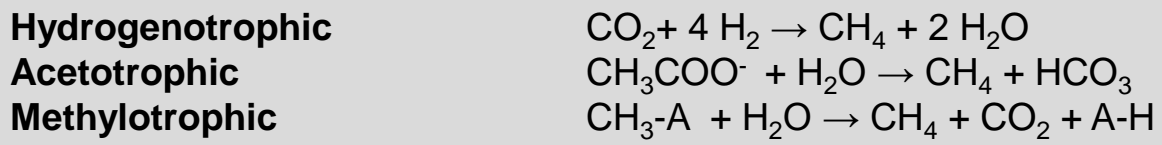


Formation of methane:

Degradation of organic matter by redox processes

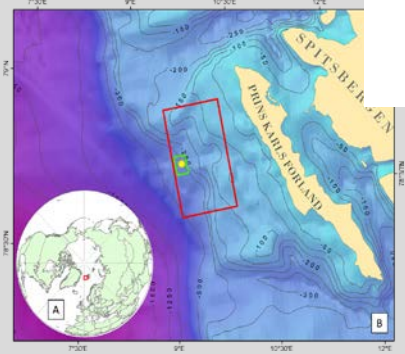
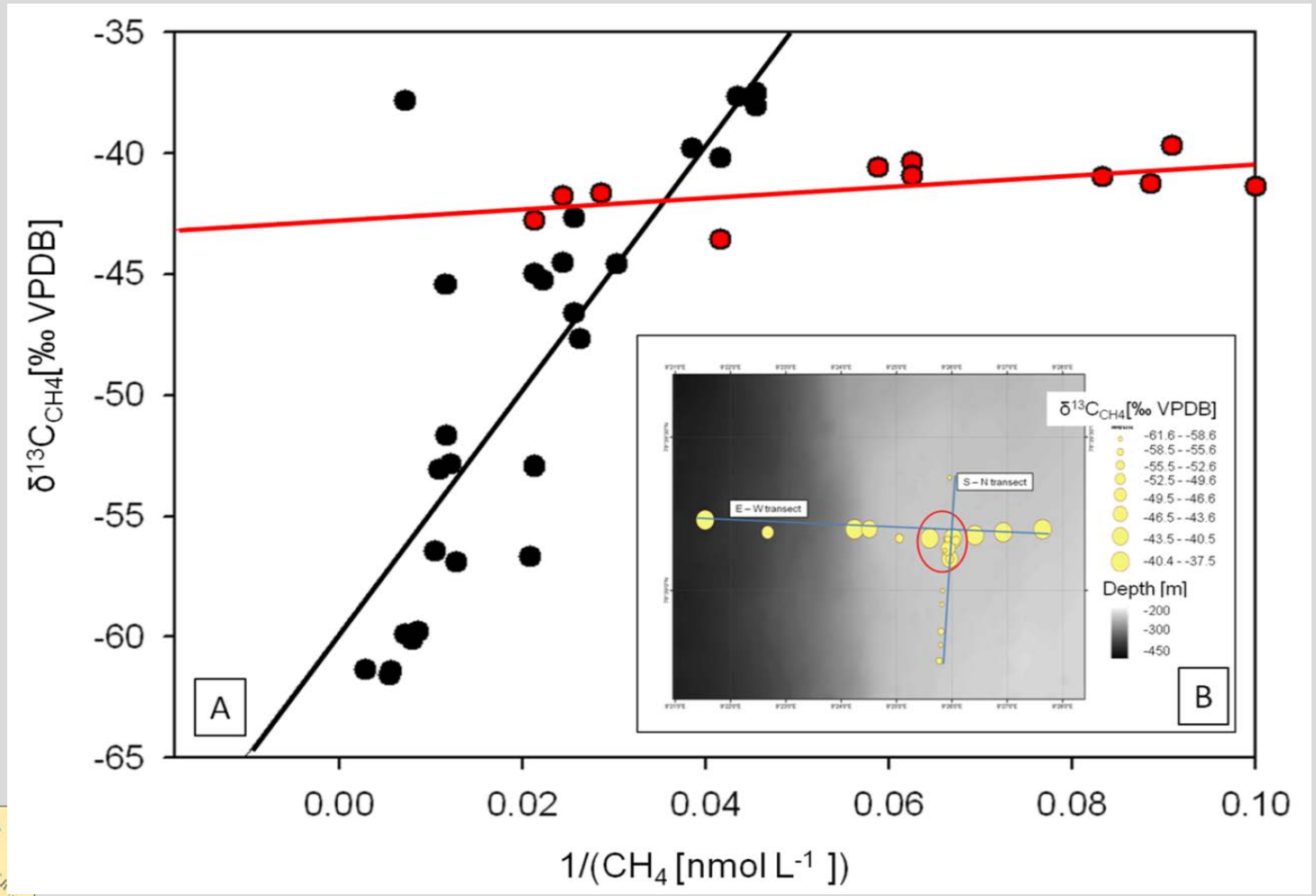


Microbial formation of methane:



Thermocatalytic formation of methane

Schematic view of the formation (modified after Froelich et al. 1979)



A) Inverse CH_4 concentration versus $\delta^{13}\text{C}_{\text{CH}_4}$ values (Keeling plot). Layer III is presented by black dots and Layer II and I by red dots.
 (B) Distribution of $\delta^{13}\text{C}_{\text{CH}_4}$ 2 m above the seafloor including the transect lines. The red circle indicates the crossing zone of the two transects

Calculation:

Bubble diameter: 7 mm by ImageJ

$$r_e = (a^2 b)^{1/3} \quad (1)$$

$$V = \frac{4}{3} \pi r_e^3 \quad (2)$$

Leifer and Patro 2002

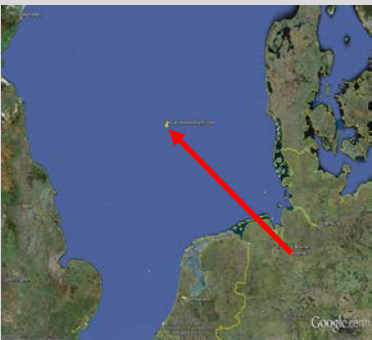
Release frequency: 23 bubbles s^{-1}

Methane flux: 28.27 L min^{-1}

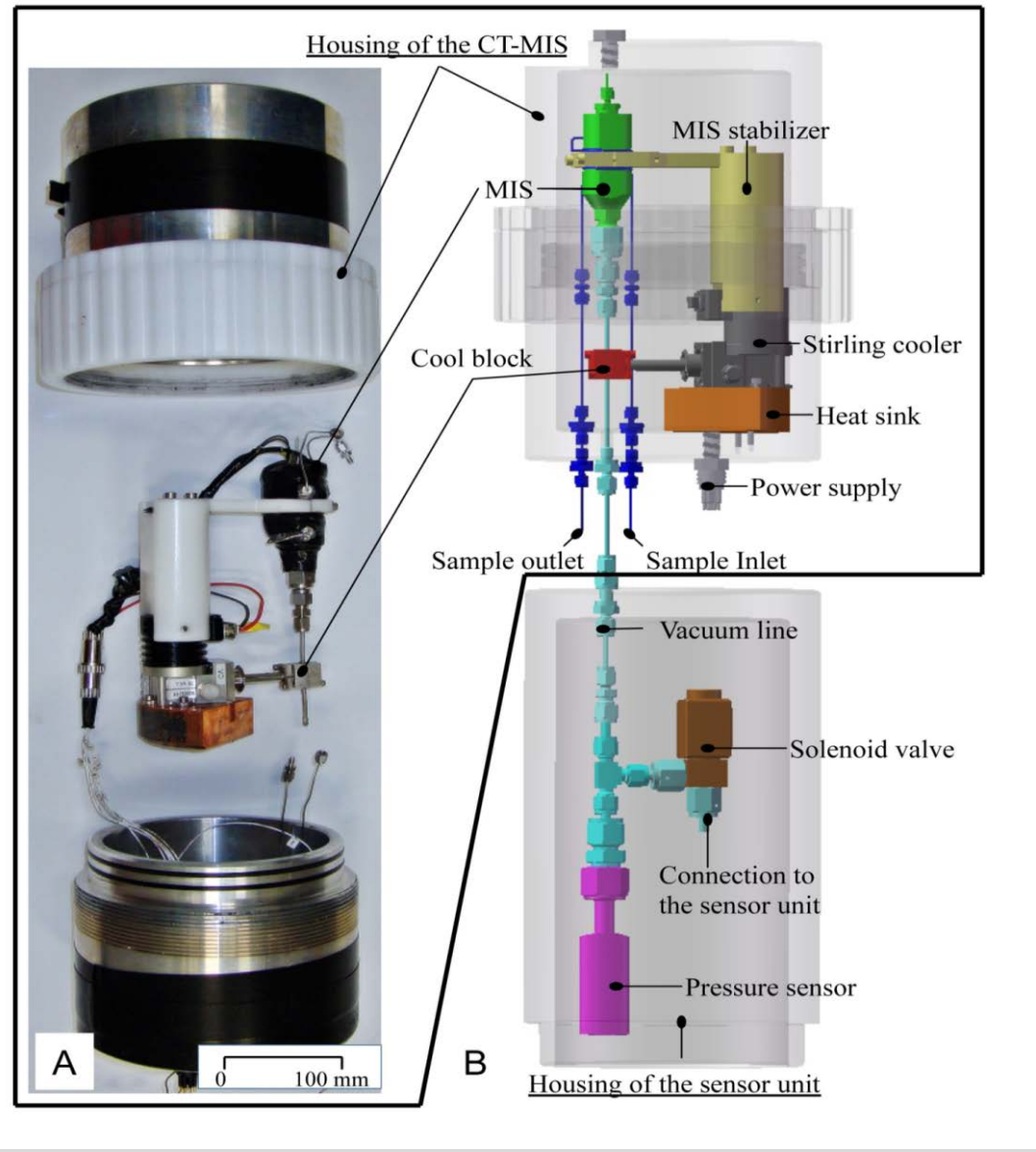
$$PVA = nRTZ \quad (3)$$

Modified after Römer et al. 2012

Seafloor methane release: 35.3 ± 17.65 t CH_4 yr^{-1}



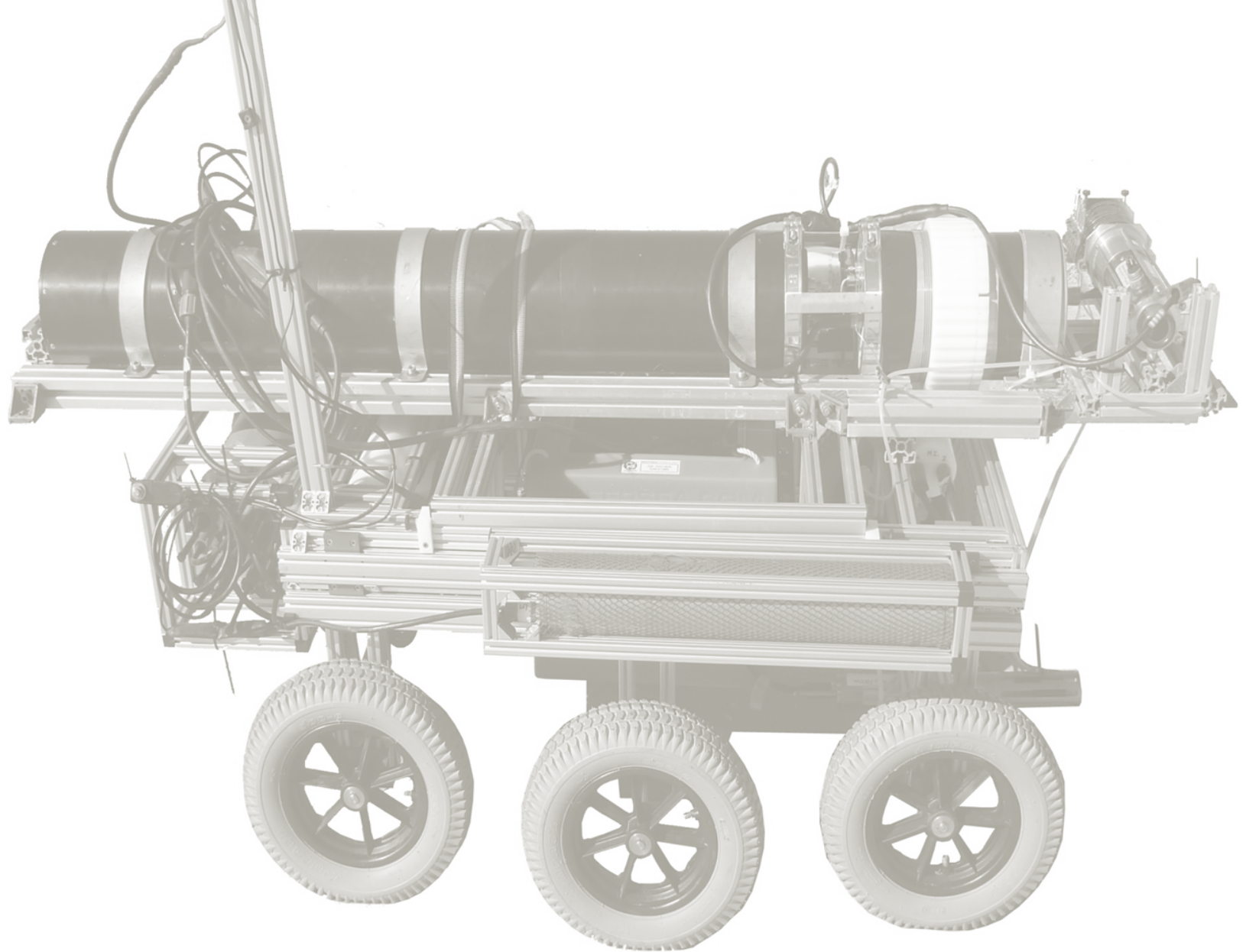
Under water cryotrap



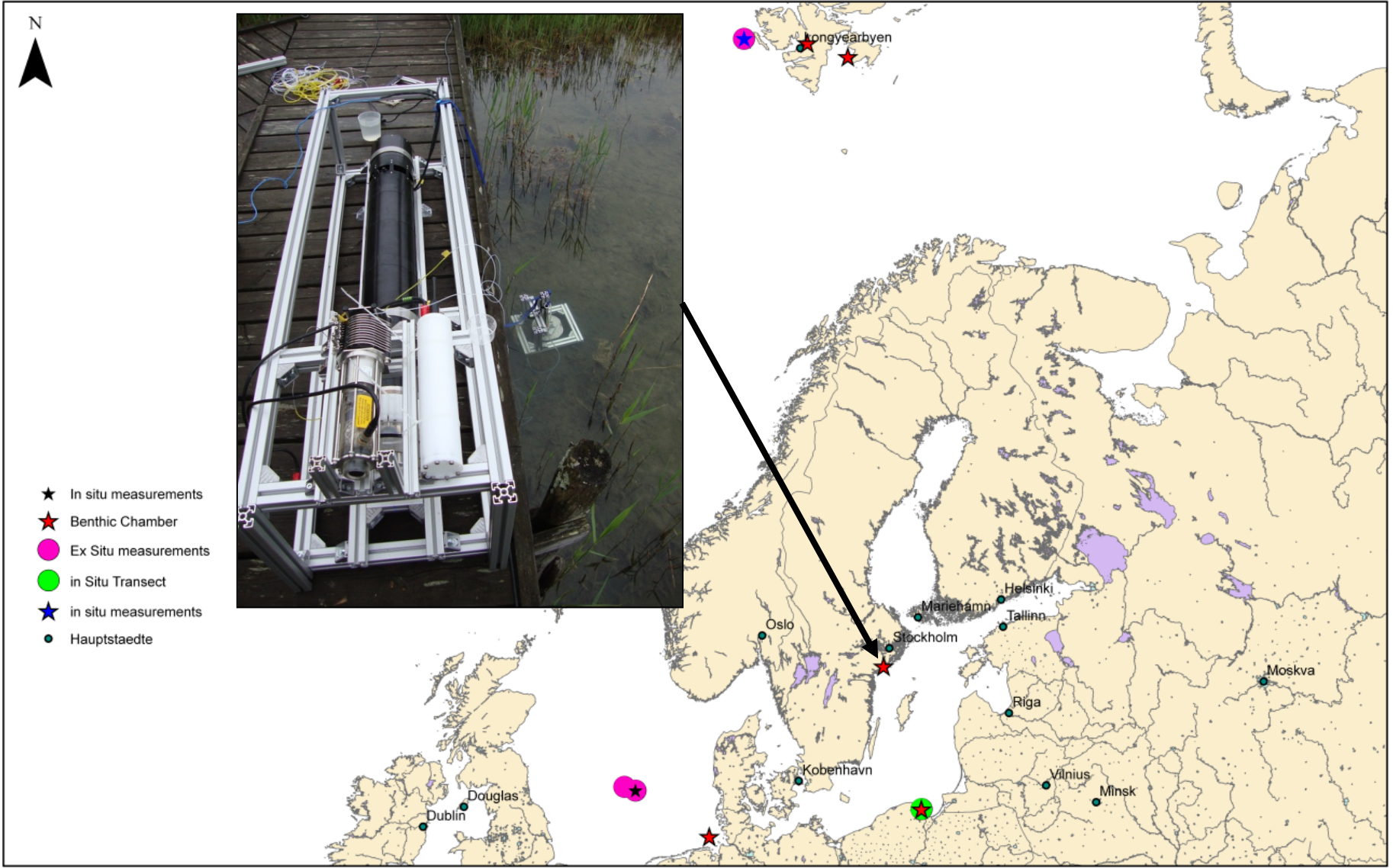
Gas analysis: New in situ sensors for high resolution mapping

TRL 1 Basic principles of technology observed and reported TRL 2 Technology concept and/or application formulated TRL 3 Analytical and laboratory studies to validate analytical predictions TRL 4 Component and/or basic sub-system technology valid in laboratory environment						TRL 5 Component and/or basic sub-system technology valid in relevant environment TRL 6 System/sub-system technology model or prototype demonstrated in relevant environment TRL 7 System technology prototype demonstrated in an operational environment TRL 8 System technology qualified through test and demonstration TRL 9 System technology qualified through successful mission operations					
Sensor	Measurement/ environments	Technology	Membrane/ Sensitive layer	Concentration range	Limit of detection	T 90	T°C	Depth range	Power supply	Manufacturer/ Research Institute/ Reference	TRL
METS- CAPSUM	Gas phase/water column	SnO ₂ semi- conductors	Silicon rubber (5–100 μm)	10 nM–150 mM	10 nM	1–30 min	2–40°C	0–3500 m	35–100 mA at 12 V	Capsium GmbH/ Franatech GmbH [26]	TRL 7
HydroC/CH ₄	Gas phase/water column	Direct IR absorption spectroscopy (3.4 μm)	Modified silicon rubber (2–100 μm)	30 nM–500 μM	<10 ppm (<6 nM)	17–30 s	0–50°C	0–6000 m	250 mA at 12 V	Contros GmbH http://www.contros.eu	TRL 7
Deep-sea methane sensor	Gas phase/water column	Laser absorption spectroscopy (3.3 μm)	Silicon-membrane tubes	40–320 ppm (25–200 nM)	40 ppm (25 nM)			0–2000 m		Hokkaido University (Japan) [15]	TRL 6/7
Deep-sea gas analyzer*	Gas phase/water column	NIR-off-axis integrated-cavity output spectroscopy	Silicon rubber			less than 1 min	0–45°C	0–2000 m	Internal battery	Iginc (USA)	TRL 6/7*
Equilibrator	Gas phase/surface water	Photoacoustic spectroscopy	Glass marbles in tube	up to 400 μM	20 μM	12 min at 7 m depth**				[33]	TRL 6
In situ mass spectrometer	Gas phase/water column	In situ mass spectrometer	Semi-permeable membrane inlet	no data	Sub-ppm (<1 nM)			0–30 m (200 m possible)	20 W	WHOI (USA) [36]	TRL 8
In situ mass spectrometer	Gas phase/water column	In situ mass spectrometer	PDMS membrane inlet	no data	1–5 ppb (<1 nM)			0–30 m (200 m possible) surface	20 W	University of South Florida (USA) [35]	TRL 8
Biosensor	Dissolved phase/sediments, pore water	Amperometry	Silicon membrane	up to 350 μM	5 μM					University of Aarhus (Denmark) [19]	TRL 5/6
Biosensor	Dissolved phase/sediments, pore water	Dissolved oxygen sensor	“bacterial beads”	0.4–2 mM	100 μM	100 s		surface		[44]	TRL 5/6
FEWS	Dissolved phase/water column	Evanescent wave spectroscopy	Optical fiber/ sensitive layer					Possibly up to 6000 m		[50]	TRL 2/3
SERS	Dissolved phase/water column	Surface-enhanced Raman scattering	Silver-colloid SERS substrate		nM–μM			Possibly up to 6000 m		Technical University Berlin (Germany) [60]	TRL 4/5
SPR	Dissolved phase/water column	Surface-plasmon resonance	PDMS/crypto- phane-A	0–400 nM	0.2 nM	2–5 min	45°C	Surface	1 mW	[64] (Appendix 2)	TRL 4/5

Compilation of in situ methane sensors and technologies, modified after Boulart (2010) including the explanation of the TRL levels, modified from a UK Defence Procurement Agency version.



Working areas



Working areas

